

May this Festival of Light add sparkles of Joy to life....

A Fortnightly Magazine Devoted to the Dyes, Dye-intermediates, Chemicals, Paints, Textile Trade & Industries



Shri B. V. Sharma
(Founder)

RANG UDHYOG NEWS



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Innovative Range of Chemicals

Pretreatment

Dyeing

Printing

Finishing



Celebrate a Green Diwali,
where every thread shines with eco-friendly light.

HAPPY DIWALI

Segments Covered

- ▶ Technical Textile
- ▶ Yarn
- ▶ Terry Towel
- ▶ Suiting
- ▶ Shirting
- ▶ Sarees
- ▶ Knits Fabrics
- ▶ Home Textile
- ▶ Denim
- ▶ Bottom Wear

Advantage of our chemicals

Less Process
Cycle

Increase
Production

Cost
Effective

GOTS
Certified
Products

Less
Pollution

Eco
friendly

Improve
Quality



Manufacturer of Textile Auxiliaries

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Pre-treatment Auxiliaries | Dyeing Auxiliaries
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ACCREDITATION
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 ISO 22000:2018
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Water Soluble Food Colours

- Tartrazine
- Sunset Yellow
- Carmoisine
- Ponceau 4R
- Amarnath
- Allura Red
- Brilliant Blue
- Indigo Carmine
- Erythrosine
- Black PN
- Red 2G
- Chocolate Brown HT
- Patent Blue V
- Quinoline Yellow

Dyes for Ink Industries

- Acid Yellow 23
- Food Yellow 3
- Acid Yellow 3
- Acid Red 18
- Acid Red 14
- Acid Red 1
- Acid Red 17
- Acid Red 51
- Acid Red 27
- Acid Blue 9
- Acid Blue 3
- Food Black 1

FD&C Colours

- FD&C Yellow 5
- FD&C Yellow 6
- FD&C Red 2
- FD&C Red 3
- FD&C Red 40
- FD&C Blue 1
- FD&C Blue 2
- FD&C Green 3

Direct Dyes

- Direct Blue 71
- Direct Blue 86
- Direct Blue 199
- Direct Red 16
- Direct Red 80
- Direct Red 81
- Direct Orange 39
- Direct Yellow 4
- Direct Yellow 86

Pigments

PIGMENT GREEN 7

- Pigment Blue 15
- Pigment Blue 15:1
- Pigment Blue 15:3
- Pigment Black

Acid Dyes

- Acid Red 52
- Acid Red 88
- Acid Red 97
- Acid Red 119
- Acid Red 131
- Acid Red 151
- Acid Red 357
- Acid Red 33
- Acid Red 73
- Acid Yellow 36
- Acid Yellow 42
- Acid Yellow 73
- Acid Blue 1
- Acid Blue 9
- Acid Blue 113
- Acid Orange 7
- Acid Violet 49
- Acid Violet 43

D&C Colours

- D&C Green 5
- D&C Green 6
- D&C Green 8
- D&C Orange 4
- D&C Red 6
- D&C Red 7
- D&C Red 17
- D&C Red 21
- D&C Red 22
- D&C Red 33
- D&C Violet 2
- D&C Ext. Violet 2
- D&C Yellow 7
- D&C Yellow 8
- D&C Yellow 10
- D&C Yellow 11

Salt Free Dyes

- Acid Red 1
- Acid Red 18
- Acid Red 33
- Acid Yellow 33
- Direct Blue 71

Also Available

- SOLVENT DYES
- BASIC DYES
- LAKE COLOURS

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- **Acid Red 249**
- **Acid Black 194**
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- **Acid Violet 54**
- **Acid Red 57**
- **Acid Yellow 110**
- **Acid Blue 113**

● Acid Yellow 2G	Yellow 17	● Acid Red RS	Red 114
● Acid Light Yellow 3GL	Yellow 34	● Acid Red Brown V	Red 119
● Acid Metanil Yellow	Yellow 36	● Acid Milling Red 3BN	Red 131
● Acid Fast Yellow MR	Yellow 42	● Acid Red F2R	Red 151
● Acid Fast Yellow E-3GL	Yellow 49	● Acid Red BRS	Red 182
● Acid Yellow 4GL	Yellow 79	● Acid Pink M	Red 186
● Acid Yellow GR	Yellow 99	● Acid Milling Red 2B	Red 249
● Acid Yellow 5GN	Yellow 110	● Acid Red S-G	Red 315
● Acid Yellow 2GL	Yellow 151	● Acid Red GN	Red 337
● Acid Yellow RL	Yellow 194	● Acid Scarlet ML	Red 357
● Acid Yellow E-4RL	Yellow 199	● Acid Red 2B	Red 361
● Acid Yellow S2G	Yellow 220	● Acid Red MBR	Red 362
● Isolan Yellow S-GL	Yellow 232	● Acid Bordeaux S-B	Red 405
● Acid Yellow E	Orange 3	● Acid Red S-3G	Red 407
● Acid Orange II	Orange 7	● Acid Red H	Red 423
● Acid Orange RO	Orange 8	● Acid Violet 4BN / 5BN	Violet 17
● Acid L. Fast Orange G	Orange 10	● Acid Brill. Red 10B	Violet 54
● Acid Orange 2G	Orange 61	● Acid Bordeaux MB	Violet 90
● Acid Orange 3RL	Orange 67	● Acid Patent Blue AS	Blue 7
● Acid Orange G	Orange 74	● Acid Brill. Blue FCF	Blue 9
● Acid Yellow 2RL	Orange 80	● Acid Blue FF	Blue 15
● Acid Orange RL	Orange 86	● Acid Blue 2AL	Blue 25
● Acid Orange	Orange 139	● Acid Blue AGG	Blue 40
● Acid Orange RL	Orange 142	● Acid Ink Blue G	Blue 93
● Acid Orange 2R	Orange 154	● Acid N. Blue S5R	Blue 113
● Acid Fast Red EG	Red 1	● Acid Blue G	Blue 158
● Acid Scarlet 3R	Red 18	● Acid Turq. Blue 8G	Blue 185
● Acid Fast Red EBL	Red 37	● Acid Dark Blue MTR	Blue 193
● Acid Red 3GP	Red 57	● Acid Blue E-4RL	Blue 277
● Acid Red G	Red 88	● Acid Blue GLE	Blue 281
● Acid Milling Red G	Red 97	● Acid N. Blue SRL	Blue 335
● Acid Brill. Red BL	Red 106		

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E-mail : jaybharat9084@gmail.com Website : www.bharatdyechem.com

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• Acid Green 3B	Green 25	• Acid Brown RL	Brown 365
• Acid Green UG	Green 73	• Acid Brown SRL	Brown 402
• Acid Olive Green BGL	Green 104	• Acid Brown LN2RL	Brown 414
• Acid Green BS	Green 111	• Acid Brown RL	Brown 425
• Acid Brown DS	Brown 58	• Acid Brown R	Brown 432
• Acid Brown ER	Brown 75	• Acid Brown 5GX	Brown 452
• Acid Brown 4BX	Brown 78	• Acid Black 10BX	Black 1
• Acid Brown CG	Brown 83	• Acid Nigrosine Black	Black 2
• Acid Brown MF	Brown 97	• Acid Black BR	Black 24
• Acid Fast Brown 5G	Brown 98	• Acid Black WA	Black 52
• Acid Orange Brown BT	Brown 106	• Acid Black BGL	Black 107
• Acid Brown NGB	Brown 161	• Acid Black BGL	Black 164
• Acid Brown NT	Brown 165	• Acid Black LDN	Black 172
• Acid Brown HH	Brown 188	• Acid Black MSRL	Black 194
• Acid Brown SGR	Brown 282	• Acid Black N	Black 210
• Acid Fast Brown ESR	Brown 348	• Acid Black NBR	Black 234
• Acid Brown ESG	Brown 349		
• Direct Yellow 2G	Yellow 11	• Direct Fast Rose FR	Red 227
• Direct F. Yellow GC	Yellow 44	• Direct Scarlet 6BS	Red 239
• Direct Yellow D-RL	Yellow 86	• Direct Violet MB	Violet 9
• Direct Yellow 3GF	Yellow 132	• Direct Helio B	Violet 51
• Direct Yellow PG	Yellow 142	• Direct Supra Violet 5BL	Violet 66
• Direct F. Orange SE	Orange 26	• Direct Sky Blue FF	Blue 15
• Direct Orange Light	Orange 34	• Direct Blue GLL	Blue 71
Fast TGLL (Y)		• Direct Fast Blue RLL	Blue 80
• Direct Orange 2GLL	Orange 37	• Direct F. Turq. Blue GL	Blue 86
• Direct Orange Light	Orange 39	• Direct Blue FBL	Blue 199
Fast TGLL (R)		• Direct Black GR	Black 22
• Direct Scarlet 4BS	Red 23	• Direct Black TOB	Black 80
• Direct Red 6BL	Red 79	• Direct Black GB	Black 168
• Direct Red 3B	Red 80	• Direct Black E-NB	Black 170
• Direct Fast Red 5BL	Red 81		

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- » ALLURA RED
- » AMARANTH
- » CHOCOLATE BROWN HT
- » QUINOLINE YELLOW W.S.
- » INDIGO CARMINE
- » ALL BLENDED COLOURS

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- » DTPTSA
- » VIOLET ACID
- » EUROPIAN K ACID
- » 2-7 DIOL
- » MIX CLEAVE ACID
- » 1-7 CLEAVE ACID
- » 1-6 CLEAVE ACID
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- » SCHEFFER'S ACID
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| Solvent Red-1 | Solvent Yellow-2 | Solvent Yellow-56 |
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| Pig. Yellow-2G | Pig. Tur. Blue-B | Pig. Bordeaux-B | Pig. Sig. Red - 112 (Ext.) |
| Pig. Gold Yellow-RM | Pig. Navy Blue-B | Pig. Rubine-B | Pig. Orange - 5 (Ext.) |
| Pig. Orange-5G | Pig. Red-GR | Pig. Red Violet-R | Pig. Yellow - 74 (Ext.) |
| Pig. Green-B | Pig. Pink-B | Pig. Olive Green-B | <input type="checkbox"/> Binder-SLN |
| Pig. Green-2G | Pig. Brown-GR | Pig. Black-R | <input type="checkbox"/> Binder-4000 |

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and
prosperous new year

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&
PROSPEROUS NEW YEAR



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"Wishing You a Bright & Joyous Diwali &
A Prosperous New Year !



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- 1-Phenyl-3-Methyl-5-Pyrazolone (CAS No. -89-25-8)
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(Cas No: 84-57-1)

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- Para amino benzoic acid
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- 2-Amino-2'4'-dichloro diphenyl ether (for Acid Yellow 110)
- 2-Amino-4-chloro diphenyl ether (for Acid Red 249)
- Amino Resorcin Di-othocresyl ether (for Acid Violet 54)

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E-mail id : marketing@technichemorganics.com
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| • REACTIVE YELLOW R (Y. 145) | • REACTIVE ORANGE 2R (O. 7) | • REACTIVE BLUE RN (PRINT) |
| • REACTIVE YELLOW H4G (Y. 18) | • REACTIVE ORANGE W3R (--) | • REACTIVE BLUE FNG (B. 187) |
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| • REACTIVE G. YELLOW HR (O. 12) | • REACTIVE RED P2B (R.45) | • REACTIVE BLACK B (B.5) |
| • REACTIVE G. YELLOW R (--) | • REACTIVE RED P8B (R.218) | • REACTIVE WNN (---) |
| • REACTIVE ORANGE H2R (O. 13) | • REACTIVE RED H4BN (R.24) | • REACTIVE MAROON R (---) |
| • REACTIVE ORANGE P2RN (--) | • REACTIVE RED BS (R.111) | |

ME DYES

- | | |
|---------------------------------|-------------------------------|
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| • REACTIVE ORANGE ME2RL (D) | • REACTIVE BLUE ME2RL (B.248) |



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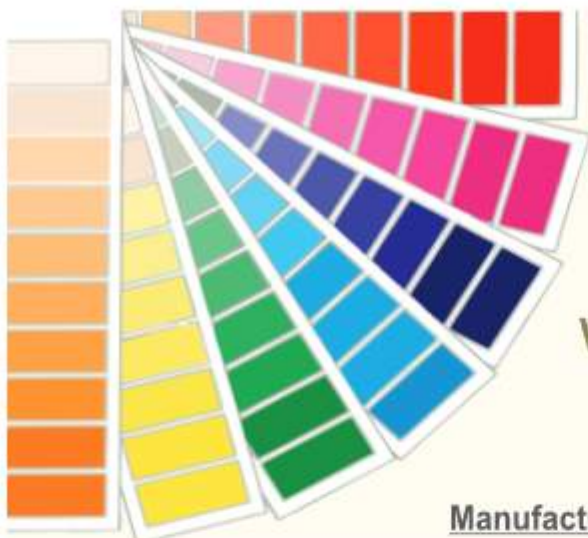
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- * Amranth
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- * Erythrosine
- * Red 2G
- * Brill. Blue FCF
- * Chocolate Brown HT
- * Indigo Blue
- * Quinoline Yellow

DIRECT DYES

- Di. Supra Yellow RL
- Di. Yellow RCH
- Di. Orange SE
- Di. Orange TGLL
- Di. Orange TGLL
- Di. Bordeaux 6B
- Di. Fast Scarlet 4BS
- Di. Fast Red BLN
- Di. Fast Red 5BL
- Di. Fast Pink CM
- Di. Fast Scarlet 6BS
- Di. Red GN
- Di. Violet BB
- Di. Blue B
- Di. Blue GLL
- Di. T. Blue GL
- Di. Black BL
- Di. Yellow 86
- Di. Yellow 99
- Di. Orange 26
- Di. Orange 34
- Di. Orange 39
- Di. Red 16
- Di. Red 23
- Di. Red 80
- Di. Red 81
- Di. Red 227
- Di. Red 239
- Di. Red 254
- Di. Violet 51
- Di. Blue 15
- Di. Blue 71
- Di. Blue 86
- Di. Black 22

ACID DYES

- Acid Yellow GR
- Acid Yellow R
- Acid Light Fast Yellow 3R
- Acid Orange II
- Acid Orange
- Acid Light Fast Orange 24
- Acid Orange 5GR
- Acid Red 2G
- Acid Crimson Red
- Acid Red 14
- Acid Scarlet 3R
- Acid Fuchine Red 2A
- Acid Red BG
- Acid Scarlet MOO
- Acid Fast Red A
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- Acid Red 18
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- Acid Red 88
- Acid Red 97
- Acid Red 119
- Acid Red 151
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* Contact Person : Mr. Rakesh Patel (M) : 09825005217

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Reactive Blue 160	Reactive Yellow H7GL
Reactive Blue 194	Reactive Yellow HER
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Contact Person : Mr. Manoj Agarwal (M) : 098250 30954 / Mr. Anurag Agarwal (M) : 098240 52301

Website : www.Radhakrishnapigments.com

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ACID RED

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- Red 5BL
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- Swiss Pink
- Rose BS
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SHRI AMBICA DYE-CHEM

L-25, G.I.D.C., B/h. Old Water Tank, Near S.B.I., Odhav, Ahmedabad - 382 415. (Gujarat-India)

E-mail : ninad.patel123@gmail.com

Contact Person : MR. CHIRAG PATEL - (M) 099092 65840
MR. MEHUL PATEI - (M) 094280 44956



CHINA INTERDYE 2026

The 25th China International Dye Industry, Pigments and Textile Chemicals Exhibition



BIZ LINK TOURS

This is 25th iteration of the China International Dye Industry, Pigments, and Textile Chemicals Exhibition, recognized as the largest specialized exhibition in the dye-chem industry globally. It's an annual event that showcases a wide range of products and technologies related to the dye industry, pigments, and textile chemicals. The exhibition includes technical programs like conferences and events held by relevant national industry associations and technical seminars by exhibitors.

JOURNEY OVERVIEW :-

DATE	DAY	SCHEDULES
14 APR	TUE	ARRIVE SHANGHAI / CHECK IN TO THE HOTEL/ DINNER
15 APR	WED	BREAKFAST / VISIT CHINA INTERDYE 2026/ DINNER
16 APR	THRU	BREAKFAST / VISIT CHINA INTERDYE 2026 / DINNER
17 APR	FRI	BREAKFAST / VISIT CHINA INTERDYE 2026 / DINNER
18 APR	SAT	CHECKOUT FROM HOTEL / DEPART FOR AIRPORT



FLIGHT DETAILS :-

SQ 505 13 APR AMD SIN 2250 0710	SQ 827 18 APR PVG SIN 0805 1325
SQ 830 14 APR SIN PVG 0930 1505	SQ 504 18 APR SIN AMD 1835 2135

HOTEL FOR THE ACCOMMODATION : SHANGHAI MOVENPICK HOTEL PEOPLE SQUARE 4** OR SIMILAR**

PACKAGE INCLUSIONS :-

- > AIRFARE WITH TAXES
- > 04 NIGHTS HOTEL ACCOMMODATION WITH 2 INDIAN DISHES IN BREAKFAST AND DINNER
- > AIRPORT - HOTEL - AIRPORT TRANSFER WITH GROUP
- > HOTEL - FAIRGROUND - HOTEL TRANSFER WITH GROUP DURING CHINA INTERDYE 2026
- > CHINA GROUP VISA
- > OVERSEAS MEDICAL INSURANCE (UPTO AGE OF 55 YEARS)
- > DINNER IN COMFORT OF HOTEL (CATERING)
- > RIVER CRUISE, SHOPPING DROP AT AP PLAZA
- > SERVICE OF BIZ LINK TOURS PROFESSIONAL TOUR MANAGER

PACKAGE EXCLUSIONS :-

- > LUNCH
- > ENTRANCE TICKETS TO FAIRGROUND
- > BEVERAGES, LAUNDRY, TELEPHONE CHARGES, PORTRAGE, EXCESS BAGGAGE & EXPENSE OF PERSONAL NATURE
- > ANY ITEM NOT MENTIONED IN PACKAGE INCLUSION

TWIN SHARING
1,19,999

SINGLE SUPPLEMENT
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EXTRAS :-

- > CHINA STICKER VISA ADDITIONAL COST 8000 INR
- > GOVERNMENT SERVICE TAX OF 5% APPLICABLE ON TOTAL PACKAGE PRICE (MANDATORY)
- > TCS - CHARGEABLE 20% ON FOREIGN CURRENCY COMPONENT OF THE PACKAGE PRICE
TCS WILL NOT BE APPLICABLE IF 2% TDS IS DEDUCTED UNDER SECTION 194/C

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KIRTI SAMTANI
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Vinyl Sulphone Based Dyes	ME Based Dyes "Bifunctional Dyes"	HE Based Dyes "Highly Exhausted Dyes"
<ul style="list-style-type: none"> • Re. G. Yellow R (Y. 201) • Re. Yellow FG (Y. 42) • Re. Orange 2R (O. 7) • Re. Yellow H7GL (Y. 57) • Re. Yellow W3R (O. 131) • Re. Red W3R (O. 37) • Re. Red B5 (R. 111) • Re. Deep Red CD (R. 278) • Re. Brill. Blue BB (B. 21) • Re. Black B (BK. 5) • Re. Black WNN (---) 	<ul style="list-style-type: none"> • Re. G. Yellow RR (Y. 145) (A) • Re. G. Yellow MERL (Y. 145) • Re. Orange ME2RL (O. 122) • Re. Yellow ME4GL (Y. 160) • Re. Yellow ME4GL (M) (Y. 186) • Re. Red ME4BL (R.195) • Re. Blue BF (B. 222) • Re. N. Blue ME2GL (B. 194) • Re. Blue RGB (B. 250) • Re. Blue ME3GL (B. 221) • Re. Green GDN (---) 	<ul style="list-style-type: none"> • Re. Red HE3B (R. 120) • Re. Red HE7B (R. 141) • Re. Red HE8B (R. 152) • Re. N. Blue HER (B. 171) • Re. Black HEBL
		M Based Dyes "Cold Dyes"
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| ● CARMOISINE (R. 3) | ● INDIGO CARMINE (B. 1) |
| ● PONCEAU 4R (R. 7) | ● BRILLIANT BLUE FCF (B. 2) |
| ● AMARANTH (R. 9) | ● QUINOLINE YELLOW (WS) (Y 13) |
| ● ERYTHROSINE (R. 14) | ● BLACK PN (Bk. 1) |



Reg Office : 317, Avani Icon, Near Haridarshan Cross Road,
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- ▶ Mono Ethanol Amine 99%
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▶ DTPTSA

▶ DTPT

▶ 3'5 DABA

▶ PABA

▶ Chromotropic Acid

▶ CABA



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- G. Yellow HER (Y.84)
- Orange HE2R (O.84)
- Red HE3B (R.120)
- Red HE7B (R.141)
- Red HE8B (R.152)
- Blue HERD (B.160)
- N. Blue HER (B.171)
- N. Blue HE2R (B.172)
- Green HE4BD (B.19)

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- Yellow ME4GL (Y.160)
- G. Yellow MERL (Y.145)
- Orange ME2RL (O.122)
- Red MERBL (R.198 A)
- Red ME3BL (R.194)
- Red ME6BL (R.196)
- Red BSRD (R.111)
- Blue BGF (B.231)
- Blue BF (B.222)
- N. Blue ME2GL (B.194)
- N. Blue ME2RL (B.248)

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- Orange H2R (O.13)
- Red 6BX (R.76)
- Red H8B (R.31)
- Red HR (R.24)
- Red H3R (R.45)
- Magenta HB (R.12)
- Purple H3R (V.1)
- Turq. Blue H5G (B.25)
- Blue HGR (B. 5)
- Blue H5R (B.13)
- Blue H3RP (B. 49)
- Red Brown H4R (B. 9)
- Black HN (Bl. 8)

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- Yellow M4G (Y.22)
- Yellow M8G (Y.86)
- Yellow MGR (Y.7)
- G. Yellow MR (Y.44)
- Yellow M4R (Y.14)
- Orange M2R (O.4)
- Red M5B (R.2)
- Red M8B (R.11)
- Magenta MB (V.13)
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- Blue M4GD (B.168)
- N. Blue M3R (B. 9)

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- Yellow R (O. 44)
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- Orange 2R (O.7)
- Orange 3R (O.16)
- Red 5B (R. 35)
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- Blue 3R (B.28)
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- Turq. Blue G (B.21)
- N. Blue GG (B. 203)
- Brown GR (Br. 18)
- Black B (Bl. 5)
- Black RL (Bl. 31)
- Black HFGR
- Black N 150



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- | | |
|-------------------------|-----------------|
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| • Red HE7B | C.I. Red 141 |
| • Red HE3B | C.I. Red 120 |
| • Red BS | C.I. Red 111 |
| • Red ME3BL | C.I. Red 194 |
| • Red ME4BL | C.I. Red 195 |
| • G. Yellow MERL | C.I. Yellow 145 |
| • Orange ME2RL | C.I. Orange 122 |
| • G. Yellow HER | C.I. Yellow 84 |
| • Orange HE2R | C.I. Orange 84 |
| • Black WNN / CNN / GDN | C.I. Black Mix |
| • Magenta Fast | |

REACTIVE COLD & HOT DYES

- | | |
|----------------|----------------|
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| • Red M8B | C.I. Red 11 |
| • Red P2B | C.I. Red 45 |
| • Magenta MB | C.I. Violet 13 |
| • Magenta HB | C.I. Magenta |
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- ★ ETHYLENE DIBROMIDE
- ★ N PROPYL BROMIDE
- ★ TRICLOSAN
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Website : www.roopdyes.com

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- D2R DYES
- GL DYES

Plot No. 439/1 & 417, Phase-II, G.I.D.C.,
Vatva, Ahmedabad - 382 445. Gujarat, INDIA.

Navin Patel : +91 - 93767 36674

(W) +91 - 079 - 4008 4648

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LOCAL & EXPORT INQUIRY IS SOLICITED



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Manufacturer and Exporter of
REACTIVE DYES

BLACKS	<ul style="list-style-type: none"> ▪ REACTIVE BLACK B ▪ REACTIVE BLACK WNW ▪ REACTIVE BLACK 39 ▪ REACTIVE BLACK PGR ▪ REACTIVE S. BLACK R / G
BLUES	<ul style="list-style-type: none"> ▪ REACTIVE BLUE FNG ▪ REACTIVE BLUE HEGN ▪ REACTIVE BLUE 221 ▪ REACTIVE BLUE 222 ▪ REACTIVE N. BLUE RGB
REDS	<ul style="list-style-type: none"> ▪ REACTIVE RED ME3BL ▪ REACTIVE RED BS ▪ REACTIVE RED MEGF ▪ REACTIVE RED BROWN H4R
YELLOWS	<ul style="list-style-type: none"> ▪ REACTIVE GOLDEN YELLOW MERL
ORANGE	<ul style="list-style-type: none"> ▪ REACTIVE ORANGE ME2RL



ISO 9001 : 2015 CERTIFIED COMPANY



- Office Address** : I / 156-A, Phase - 2, G.I.D.C., Vatva, Ahmedabad - 382445. (Gujarat) India.
- Factory Address** : Saykha
- Email** : info@lioncolor.com, devlioncolor@gmail.com
- Web** : www.lioncolor.com
- Contact** : +91 - 93270 02743
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✓ Resist Salt Powder	127-68-4	✓ PAABSA	104-23-4
✓ Resist Salt Granular	127-68-4	✓ NN DMA	121-69-7
✓ Metanilic Acid	121-47-1	✓ DPA (Diphenylalanine)	122-39-4
✓ Aniline 2:4 DSA	137-51-9	✓ OAVS	10079-20-6
✓ Aniline 2:5 DSA	98-44-2	✓ Vinulsulphone (VS)	2494-89-5
✓ B.D.S.A.	117-61-3	✓ Sulpho OAVS	121-88-0
✓ DMAP	100-17-4	✓ DMAVS	26672-24-2
✓ MAP	1122-58-3	✓ PCVS	21635-69-8
✓ Hydrazobenzene	122-66-7	✓ PNCB	100-00-5
✓ MPD	108-45-2	✓ PNCBOSA	946-30-5
✓ MPDSA	88-63-1	✓ 2:4 DNCB	97-00-7
✓ PNT	99-99-0	✓ 2:5 DCNB	89-61-2
✓ PNTOSA	121-03-9	✓ MCB	108-90-7
✓ DNSDA	128-42-7	✓ PNA	100-17-4
✓ DASDA	81-11-8	✓ PPD (Lumps / Powder)	106-50-3
✓ ONT	88-72-2	✓ ONP (Ortho-Nitrophenol)	88-75-5
✓ 4 Sulfo Anthranilic Acid	98-43-1	✓ PNP (Para-Nitrophenol)	100-02-7
✓ 4 Sulfohydrazone	118969-29-2	✓ Chloranil	118-75-2
✓ Aniline Oil	62-53-3		

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✓ Beta Naphthol	135-19-3
✓ G Salt	118-32-1
✓ R Salt	148-75-4
✓ K Acid	118-03-6
✓ Gamma Acid	90-51-7
✓ Tobias Acid	81-16-3
✓ STA	117-62-4
✓ J Acid Urea	34-47-4
✓ J Acid	87-02-5
✓ Sulpho Gamma Acid	90-40-4
✓ Sulpho J Acid	6535-70-2

BASIC CHEMICALS

✓ Soda Ash	497-19-8
✓ Sodium Bicarbonate	144-55-8
✓ Sodium Nitrite	7632-00-0
✓ Sodium Bisulphite	7631-90-5

Contact Person : MR. SANJAY PANCHAL (M) +91- 98795 88391

Sister Concern :- **NI Nine International**

Plot No. 117, Phase-II, Opp. Meghmani Dyes & Intermediates, G.I.D.C. Vatva, Ahmedabad - 382 445, (Gujarat) INDIA.
E-mail : nineinternational3@gmail.com



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- Orange G (74)
- Black WA (52)
- Blue 2G (158)
- Pink BE (Red 186)
- Red 2G (Red 183)
- Violet Br.B (Red 184)

Leather Browns

- Brown 75
- Brown 97
- Brown 98
- Brown 165

Milling Dyes

- Red 119
- Blue 113
- Black 24
- Black 26

1:2 Metal Complex Dyes

- | | |
|------------------------|----------------------|
| • Yellow M3RL (194) | • Bordeaux MB (90) |
| • Orange M2RL (142) | • Yellow M2GLN (241) |
| • Brown MGRL (402) | • Acid Green (104) |
| • Brown MBL (355) | • Gray SBL (-) |
| • Brown M2RL (365) | • Black 107 (-) |
| • Acid Brown (282) | • Black GR (172) |
| • Scarlet ML (Red 357) | • Black MSRL (194) |
| • Red MBR (Red 362) | |
| • Blue MTR (193) | |



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- Blue EDG
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- Blue 221
- Blue 220
- Blue 250
- Red 223 – META & PARA
- Black PGR
- Blue P2R (Black 39)
- Red 31
- Red 24
- Red 45
- Red 3:1
- Brown 11
- & All Printing Reactive Dyes

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- RED GL SALT
- RED RL SALT
- RED TR SALT
- RED 3GL SALT
- SCARLET R SALT
- SCARLET GGS SALT
- BORDEAUX GP SALT
- BLUE B SALT
- BLUE BB SALT
- YELLOW GC SALT
- ORANGE RD SALT
- BLACK B SALT
- BLACK K SALT

DYCROMINE Sunfast Direct Dyes

- Black - 22
- Blue - 15, 71, 151, 201
- Yellow - 86, 142
- Red - 23, 31, 227
- Orange - 26, 39
- Violet - 9, 51

REACTIVE BLACKS

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 Unit-2 : 415/2/2, Phase-II, G.I.D.C., Vatva, Ahmedabad - 382 445. (INDIA)
 Tel. : +91 - 79 - 29708400 Mobile + 91 - 9879484214
 E-mail : info@jagson.com / exports@jagson.com
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Phone : +91-90999 77210

E-mail : exports@धारप्रतिकशा.com

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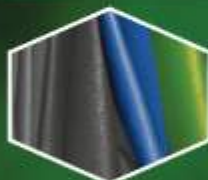
Factory : Plot No. DP-24, Saykha Industrial Estate, GIDC Saykha, Road No. 30, Tal. Vagra, Dist. Bharuch-392140. Gujarat, INDIA

Contact Person : Mr. Jayesh Patel (Director) (M) +91 - 93761 06071

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- **ACID BLUE 113**
- **ACID RED 119**
- **ACID BROWN 354**
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- BLACK MSRL**
- BLACK NT**
- BLACK NB**
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• E-mail : poojachemind@gmail.com • Website : www.poojachemind.com

Contact Person : MR. RAHUL MALANI (M) 099248 83156

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Ta. Kalol, Dist. Gandhinagar (North Gujarat), INDIA.

E-mail :- pbhagwatiindustries112@gmail.com Website : www.bhagwatiindustries.com

Contact Person : Mr. Varun Patel (M) +91 - 81560 01293 / Bhagwanbhai Patel (M) +91 - 98250 15565 /
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• RE. ORANGE HE2R (ORANGE 84)

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- Re. Orange ME2RL (Orange 122)
- Re. Red M5B (Red 2)
- Re. Red HE7B (Red 141)
- Re. Red HE8B (Red 152)
- Re. Magenta MB (Violet 13)
- Re. Blue MR (Blue 4)
- Re. Blue M2R (Blue 81)
- Re. Navy Blue M3R (Blue 9)
- Re. Navy Blue RX (Blue 59)
- Re. Blue ME2RL (Blue 248)
- Re. Green HE4BD (Green 19)

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- Re. Navy Blue HE2R (Blue 172)
- Re. Blue HERD (Blue 160)



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Mr. VIJAY SINGWALA :- (M) 09377131456 , 09429012980 / Mr. DILOZ SINGWALA (M) 09904531456

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- METHYL VIOLET (Powder/Crystal/Liquid)
- MALACHITE GREEN (Powder/Crystal/Liquid)
- CRYSTAL VIOLET (Liquid)
- BISMARCK BROWN - Y / R (Powder/Liquid)
- RHODAMINE B (Powder/Liquid)
- ETHYL VIOLET (Liquid)
- VICTORIA BLUE (Powder/Liquid)
- MAGENTA (Powder)

ACID DYES

- METANIL YELLOW (Yellow 36)
- ORANGE II (Orange 7)
- SCARLET 3R (Red 18)
- SCARLET MOO (Red 73)
- RED 3BN (Red 131)

VAT DYES

- VAT PINK R

FAST SALT

- FAST RED B BASE
- FAST RED RC

PIGMENT POWDER

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- ORANGE 13

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:- Sister Concern :-



Mahendra Dyes and Chemicals

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Phone : +91 - 02646 - 221503 E-mail : singhwala@gmail.com Website : www.harishchemiclas.com
Contact : Mr. Vijay Singhal - (M) 09377131456 / Mr. Mukesh Mehta (M) 09327908059



Manish Industries

Factory : Plot No. 3921, Behind Harish Chemicals, G.I.D.C. Ankleshwar-392 002. (Gujarat) INDIA
Phone : +91 - 02646 - 223094 E-mail : singhwala@gmail.com Website : www.harishchemiclas.com
Contact : Mr. Vijay Singhal - (M) 09377131456 / Mr. Mukesh Mehta (M) 09327908059



HARISH CHEMICALS ENGG. ENTERPRISE

Factory : Plot No. 3903, C/1-B, G.I.D.C., Ankleshwar - 393 002.
Phone : +91 - 0220350 , 225397 E-mail : singhwala@gmail.com Website : www.harishchemiclas.com
Contact : Mr. Vijay Singhal - (M) 09377131456 / Mr. Mukesh Mehta (M) 09327908059



Plot No. - 769/7, Nr. Chaitanya Life Science, G.I.D.C., Jhagadia - 393110. Dist-Bharuch
Email:- info@modheshwari.co.in
Website:- www.modheshwari.co.in
Contact :- Mr. Diloz Singwala - 09904531456

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BASIC RHODAMINE B BASE



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E-mail : karmachemical@gmail.com , tankishan5115@gmail.com



Contact Persons : Mr. Prafulchandra Tank (Bsc. , HONS) (M) +91- 98250 53258
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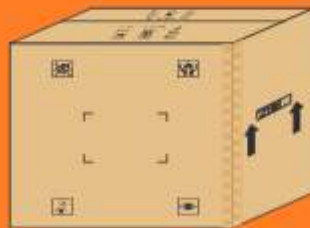
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• RE. RED P2B	(R. 45)	• RE. ORANGE W3R	(---)
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• RE. DEEP RED CD	(R. 278)	• RE. VIOLET 5R	(Vi. 5)
• RE. RED W3R	(---)	• RE. NAVY BLUE RGB	(---)
• RE. MAROON R	(---)		
• RE. G. YELLOW R(R.O. CRUDE)	(Y. 90)		
• RE. G. YELLOW RR	(---)		
• RE. YELLOW W3R	(---)		
• RE. YELLOW ME4GL	(Y. 160)		

REACTIVE BLACKS

- RE. BLACK B (HIGH CRUDE (Bk. 5)
- RE. BLACK WNN (---)

ALL MIX BLACKS

MATCHING SHADES

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- Re. Navy Blue RX (Blue 59)
- Re. Blue ME2GL (Blue 194)
- Re. Blue ME2RL (Blue 248)
- Re. Yellow ME4GL (Yellow 160)
- Re. Yellow MERL (Yellow 145)
- Re. Orange ME2RL (Orange 122)
- Re. Red ME4BL (Red 195)
- Re. Green HE4BD (Green 19A)
- Re. Navy Blue HER (Blue 171)
- Re. Black HEBL (Black 180)
- Re. Yellow HE4G (Yellow 81)
- Re. Yellow HE6G (Yellow 135)
- Re. Red HE8B (Red 152)
- Re. Orange HE2R (Orange 94)
- Re. Red HE3B (Red 120)
- Re. Red HE7B (Red 141)
- Re. Green Special (---

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- Re. Blue GG (Blue 203)
- Re. Blue BB (Blue 220)
- Re. Blue MEBF (Blue 222)
- Re. Blue RGB (Blue 250)
- Re. Blue BRF (Blue 221)
- Re. Violet 5R (Violet 5)
- Re. Red BSID (Red 111)
- Re. Black B (Black 5)
- Re. Brown GR (Brown 18)
- Re. Red W3R (---
- Re. Orange W3R (---
- Re. Red CD (---
- Re. Wine Blue (---
- Re. Coffee Brown (---
- Re. Marron XL (---
- Re. Blood Red (---

- Re. Magenta MB (Violet 12)
- Re. Violet C4R (Violet 13)
- Re. Blue MR (Blue 4)
- Re. Blue M2R (Blue 81)
- Re. N. Blue M3R (Blue 9)
- Re. Red M5B (Red 2)
- Re. Red M8G (Red 11)
- Re. Orange M2R (Orange 4)
- Re. Yellow M8G (Yellow 86)
- Re. Yellow M4G (Yellow 22)
- Re. Yellow MGR (Yellow 7)
- Re. Yellow M4R (Orange 14)
- Re. Blue M4GD (Blue 109)
- Re. Yellow P6G (Yellow 95)
- Re. Yellow RR (---
- Re. Magenta HB (Violet 26)
- Re. N. Blue HE2R (Blue 172)

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- Di. Sky Blue FF (Blue 15)
- Di. Red 5BL (Red 81)
- Di. Turq. Blue (Blue 86)
- Di. Yellow 5GLL (Yellow 44)
- Di. Viscose Orange (Orange 108)
- Di. Helio B (Violet 51)



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NON METALLIC



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AOD-PF-300



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AOD-PF-400



1 1/4" BSP (40mm) POLYPROPYLENE, PVDF, ETFE
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- Red 52

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- Yellow 16
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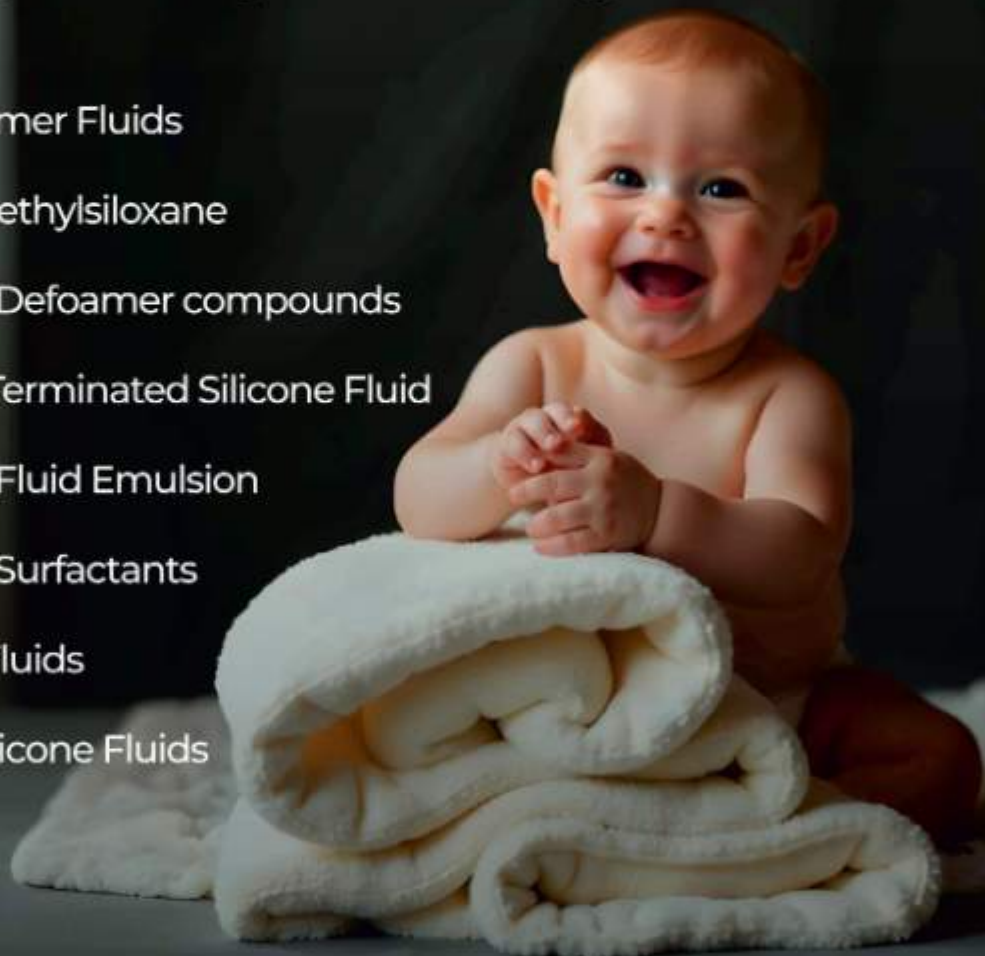
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- YELLOW 59
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RANG UDHYOG NEWS

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E-mail : arihant29277@yahoo.com , sdipak_malay@yahoo.com



Contact Person : MR. MALAY SHAH (M) +91 - 98251 91992

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-



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CONTACT PERSON : MR. MALAV S PATEL (M) 098250 64889



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E-mail : sudamakharya1990@gmail.com

Contact Person : SUDAMAPRASAD KHARYA (M) 09450153769, 08318104099



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- **Re. Yellow 95**
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- **Re. Yellow HE6G (135)**
- **Re. Blue ME2GL (194)**
- **Re. Blue ME2RL (248)**



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E-mail : rangtex_industries@yahoo.com

Contact : Mr. Hansraj Dudhat (M) 098250 43066



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- **Di. Violet MB** (Vi. 9)
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- **Di. Red 12B** (R. 31)
- **Di. Bourdeaux BW** (---)
- **Di. Brown MR** (Br. 2)
- **Di. Black E** (---)

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E-mail : siindustries.1991@gmail.com

Contact Person : MR. KETAN PATEL (M) 095371 65523 / MR. BHARAT PATEL (M) 098256 99678

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- Navy Blue HER (Blue 171)
- Navy Blue HE2R (Blue 172)
- Blue HEGN (Blue 198)
- Blue BB (Blue 220)

DIRECT DYES

- ★ **BLUE GLL** (Blue 71)
- ★ **SKY BLUE FF** (Blue 15)
- Red BLN (Red 80)
- Red 5BL (Red 81)
- Orange TGLL (Orange 34)

ACID DYES

- ★ **MAROON V** (Red 119)
- ★ **BLUE S5R** (Blue 113)
- Scarlet G (Red 97)
- Red F2R (Red 151)
- Orange II (Orange 7)

Plot No. 131/2, Phase 2, Modern Bakery Road, Opp. Lalji Maulji Transport, G.I.D.C.,
Naroda, Ahmedabad-380025. Gujarat, INDIA.

Phone : (F) +91 - 079 - 2280 1034 E-mail : shreeji320@gmail.com

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TEXTILE ENZYMES:- Desizing, Bio-Polishing, Bio-Scouring, Peroxide-Killer, etc. ALL MNC EQUIV. TEXT. AUXI. (ECO-STD).

TEXTILE CHEMICALS :- Acid Slurry, Sodium Sulfide Sol. Acetic substitute, Sodium Acetate, Descaling Agent, Water Softner, Liq. Soap, Detergent Powder, Meta Claw CL. DTPA, Ecosoft-100/SH-20, Securan-540, Peroxide Killer (CIBA STD)

ORGANIC PIGMENTS :- Full Range Powder-Latest Formulas MNC-STD.SSI/EOU Units Formulas, With Surface Treatment Procedures. [Files Available].

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PAPER CHEMICALS :- Paper Defoamer [KS/KFS/KF-229 EYC.], DSR-RESIN, POLY-DADMAC.

**WOOL, NYLON, POLYAMIDE
1. DFA & RESERVING-MR
2. ANTI PHOTO-BLG, ABP**

ORGANIC PEROXIDES : BP [POWDER & PASTE], MEKP, TBHP, TBPB, LP. etc.

OIL TECHNOLOGY :- Antistatic Oil, LV-40/52, WBO, Weight Increasing Spray-Oil for Grey Cloth, Warming Oils, Machine Oil, Loom Oil, Cutting Oil, Purification of JBO / LDO, Anti Dusting Oil, CSO, SPRAY DRIER OIL SD-1000 [ECO-FRIENDLY & LOW-COST]

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- > **METANIL YELLOW** (Yellow 36)
- > **YELLOW 5GN** (Yellow 110)
- > **SCARLET 3R** (Red 18)
- > **SCARLET M00** (Red 73)
- > **FAST RED A** (Red 88)

BASIC DYES

- > **BISMARK BROWN Y** (Brown 1)
(POWDER & LIQUID)
- > **BISMARK BROWN R** (Brown 4)
- > **AURAMINE O** (Yellow 2)
- > **CHRYSODINE R** (Orange 1)
- > **CHRYSODINE Y** (Orange. 2)
- > **MALACHITE GREEN** (Green 4)
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- * Di. Red 12B - Red 31
- * Di. Viscose Orange BRO - Orange 108

Plot No. I-312/1, Phase-II, G.I.D.C. Estate, Vatva, Ahmedabad -382 445, (Gujarat) INDIA.
E-mail : maulik8930@yahoo.com

Contact : MR. MAULIK PATEL (M) +91-97279 22444

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60, Pushpak Industrial Estate, Phase-1, G.I.D.C., Vatva, Ahmedabad - 382 445.
E-mail : zankitpatel1997@gmail.com

Contact Person : MR. MAHESH PATEL (M) : +91 - 99090 20903
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- MALACHITE GREEN
- AURAMINE O
- BISMARCK BROWN R
- BASIC MAGENTA

LIQUID FORM

- RHODAMINE B LIQUID
- METHYL VIOLET LIQUID
- MALACHITE GREEN LIQUID
- CRYSTAL VIOLET LIQUID
- BISMARCK BROWN LIQUID
- BRILLIANT GREEN LIQUID

SOLVENT DYES

- SOLVENT RED 49
- SOLVENT VIOLET 8
- SOLVENT BLACK 46

ACID DYES

- METANIL YELLOW
- ACID ORANGE II
- ACID SCARLET 3R

Contact Manufacturers :

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- ★ Methanesulfonyl Chloride - 124-63-0

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- ★ Phthalic Anhydride - 85-44-90
- ★ Succinic Anhydride - 108-30-5
- ★ Maleic Anhydride - 108-31-6

Anilines

- ★ Aniline - 62-53-3
- ★ N,N-di-ethyl-Aniline - 91-66-7
- ★ P-anisidine - 104-94-9
- ★ N,N-Dimethylaniline - 121-69-7

Aromatic

- ★ 2-Chlorobenzaldehyde - 89-98-5
- ★ 4-Chlorobenzaldehyde - 104-88-1

Hydroxylamine Salts

- ★ Hydroxylamine Hydrochloride - 5470-11-1

Inorganics

- ★ Aluminium Chlorides Anhydrous - 7446-70-0
- ★ Sodium Amide - 77820-92-5
- ★ Potassium Iodide - 7681-11-0
- ★ Sodium Bromate - 7789-38-0
- ★ Sodium Hydride - 7646-69-7
- ★ Sodium Iodide - 7681-82-5
- ★ Sodium Metaperiodate - 7790-28-5
- ★ Zinc Oxide 1314-13-2
- ★ Sodium Metal 7440-23-6

Phenols

- ★ 1-Naphthol - 90-15-3
- ★ 2-Aminophenol - 95-55-6
- ★ 2-Methyl Resorcinol - 608-25-3
- ★ 2-Naphthol - 135-19-3
- ★ 3-Aminophenol - 591-27-5
- ★ 4-Aminophenol - 123-30-8
- ★ 2-Chlorophenol - 95-57-8

Quaternary Ammonium Salts

- ★ Tetrabutylammonium Bromide - 1643-19-2
- ★ Tetramethylammonium Chloride - 75-57-0
- ★ Tri-Ethyl-Benzyl-Ammonium Chloride - 56-37-1

Reducing Agents

- ★ Sodium m-Nitro-Benzenesulfonic Acid - 127-68-4
- ★ m-Nitrobenzoic Acid - 121-92-6
- ★ Paraformaldehyde - 30525-89-4
- ★ Monochloroacetic acid - 79-11-8
- ★ 3-Nitrobenzaldehyde - 99-61-6
- ★ 2-Nitrobenzaldehyde - 552-89-6

Dyes

- ★ Reactive Dyes (P and H) Dyes
- ★ Black HN (Black 8)
- ★ Black 39
- ★ Blue 49
- ★ Blue 13
- ★ Orange 12
- ★ Purple H-3 (Violet 1)



**AAKAR
DYE**

For complete list, please contact us

Mr. BHARAT R. MEHTA

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- Reactive Red 111
- Reactive Red 198
- Reactive Blue 19
- Reactive Blue 21
- Reactive Blue 220
- Reactive Blue 221
- Reactive Blue 250
- Reactive Black WNN
- Reactive Black N-150
- Reactive Super Black G
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The year 2025 is on the verge of its completion and with these completion of a fantastic year the Indian new year is about to shine with its shining colours.

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GEMSTONE DYEING - PROCESS, ETHICS & DETECTION



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The practice of dyeing gemstones represents a significant intersection of artistry and chemistry in the jewellery world. This enhancement technique, while centuries old, continues to evolve with modern technology, offering both opportunities and challenges in the gem trade. Understanding gemstone dyeing is crucial for anyone interested in jewelry, whether as a consumer, collector, or professional, as it affects everything from a stone's value to its care requirements.

What is Gemstone Dyeing ?

Gemstone dyeing is a treatment process that has been used for thousands of years to enhance or alter the natural color of stones. While ancient civilizations used primitive methods like soaking stones in natural dyes, modern technology has revolutionized these techniques, allowing for more permanent and sophisticated color modifications. This process primarily targets porous gemstones or those with surface-reaching fractures, transforming otherwise fewer desirable stones into vibrant pieces suitable for jewelry.

* The traditional soaking method involves immersing stones in colored solutions for extended periods, sometimes lasting weeks or months, allowing the dye to naturally penetrate the stone's structure through its pores and fissures. This technique remains common for treating agates, chalcedony, and other porous stones, though results can vary in permanence and uniformity.

* Vacuum-pressure impregnation represents the cutting edge of gemstone dyeing technology, using specialized equipment to force dyes deep into the stone's structure under controlled pressure and temperature conditions. This modern approach typically produces more stable and consistent results than traditional methods, though it requires significant technical expertise and specialized equipment.

* The chemistry behind gemstone dyes has evolved from simple organic compounds to complex synthetic solutions specifically engineered for different types of stones. These modern dyes are designed to resist fading from light exposure and to maintain their color even after prolonged wear, though their effectiveness still largely depends on the stone's natural characteristics.

* High-pressure treatment systems can combine dyes with polymers or other stabilizing agents, creating a more permanent enhancement that also helps strengthen the stone. This advanced technique is particularly valuable for treating stones with surface-reaching fractures, as it can both enhance color and improve durability.

* Some stones undergo multiple dyeing processes to achieve specific colors or effects. For instance, black onyx is typically created by treating chalcedony first with a sugar solution

that is then carbonized through acid treatment, creating a permanent black color that penetrates the entire stone.

* The effectiveness and permanence of gemstone dyeing largely depend on the stone's natural porosity, the presence of surface-reaching fractures, and its chemical composition. Stones like jade, turquoise, and pearls are frequently dyed because their natural structure allows for successful color enhancement, while dense stones like diamonds and sapphires cannot be effectively dyed.

The development of gemstone dyeing techniques continues to advance, offering new possibilities for enhancing fewer valuable stones while making colored gems more accessible to a wider market. While these treatments significantly affect a stone's value compared to untreated specimens, they represent an important segment of the jewelry industry that, when properly disclosed and understood, provides valuable options for both manufacturers and consumers. The key to working with dyed stones lies in understanding their nature, limitations, and care requirements, ensuring that both retailers and customers can make informed decisions about their jewelry purchases.

What Gemstones Are Usually Dyed ?



The practice of gemstone dyeing isn't applied universally across all stone types, but rather focuses on specific varieties that are naturally receptive to color enhancement. Understanding which stones are commonly dyed helps buyers make informed decisions and sets appropriate expectations for both appearance and value. The susceptibility to dyeing largely depends on a stone's natural porosity, structure, and the presence of surface-reaching features.

* **Agate** and Chalcedony are among the most frequently dyed gemstones due to their naturally porous structure. These stones can accept dye readily and evenly, allowing for creation of vibrant colors like hot pink, electric blue, or deep purple that don't occur naturally. The treatment is typically very stable in these materials due to their microcrystalline structure.

* **Jade** (particularly nephrite and some jadeite varieties) often undergoes dyeing to enhance its color or create more desirable green shades. The process typically targets stones with lower transparency or less desirable natural colors. Natural jade has microscopic interlocking crystal structures that can accept dye when proper techniques are applied.

* **Pearls**, especially freshwater varieties, are commonly dyed to create exotic colors or to improve uniformity in matched strands. The organic nature of pearl nacre allows for effective dye absorption, though the treatment's durability can vary significantly based on application method and pearl quality.

* **Turquoise** frequently undergoes both dyeing and stabilization treatments. The stone's naturally porous structure makes it ideal for enhancement, though high-quality natural specimens are rarely treated. The treatment often combines color enhancement with stabilizing agents to improve durability.

* **Coral**, being an organic gem material, readily accepts dyes. The practice is common for creating more intense red colors or unusual shades like black or blue. The natural porosity of coral makes it particularly receptive to dye treatments.

* **Quartz** varieties, including crackled quartz and some crystalline specimens, can be dyed when they contain surface-reaching fractures or have been deliberately treated to

accept color. This includes creating popular varieties like “cherry quartz” or other artificially colored varieties.

* **Onyx** technically a variety of chalcedony, is almost always created through dyeing. Natural **black onyx** is extremely rare; most commercial black onyx is actually dyed chalcedony that has undergone specific chemical treatments to achieve its deep black color.

* **Beryl varieties** (excluding emerald and aquamarine) with lower quality or less desirable colors sometimes undergo dyeing, particularly when they contain surface-reaching fractures that can accept color enhancement.

The prevalence of dyed gemstones in the market reflects both consumer demand for affordable colored stones and the industry’s ability to enhance fewer valuable materials. While these treatments significantly affect a stone’s value compared to natural specimens, they serve an important role in making colorful jewelry accessible to a wider market. The key to working with dyed stones lies in proper disclosure, understanding their care requirements, and setting appropriate expectations for their longevity and value. Whether **natural or enhanced**, each stone has its place in the jewelry market, provided that treatments are properly disclosed and understood by all parties involved.

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Methods Used in Gemstone Dyeing

The techniques used to dye gemstones have evolved significantly over time, ranging from simple soaking methods to complex technological processes. Each method is chosen based on the stone's characteristics, desired outcome, and intended use in jewelry. Understanding these methods helps both professionals and consumers appreciate the complexity of the treatment and its implications for the stone's durability and care.

* Traditional soaking methods involve immersing stones in dye solutions for extended periods, sometimes lasting several weeks. The process requires careful monitoring of temperature, solution concentration, and exposure time to achieve optimal results. This method is particularly effective for highly porous stones like agate and can achieve deep, lasting color when properly executed.

* Pressure dyeing represents a significant advancement in gemstone enhancement technology. This method uses specialized equipment to force dye into the stone's structure under controlled pressure conditions, typically ranging from 10,000 to 40,000 psi. The high pressure enables deeper dye penetration and more uniform color distribution than traditional soaking methods.

* The polymer-dye combination treatment involves introducing colored polymer solutions into the stone's structure. The polymer helps stabilize the dye and can fill microscopic fissures, potentially improving both the stone's appearance and durability. This method is particularly valuable for treating stones with surface-reaching fractures or those that might be too fragile for high-pressure treatments.

* Surface diffusion treatment, while technically a form of dyeing, uses heat and chemicals to create a thin layer of artificial color on the stone's surface. This method is less common but can be effective for certain types of stones where deep penetration isn't necessary or desired. The results are typically less permanent than other dyeing methods.

* Fracture filling with colored substances combines repair and enhancement techniques. This method specifically targets surface-reaching cracks and fissures, filling them with colored resins or oils that both mask the fractures and add color to the stone. While

effective for improving appearance, these treatments may require periodic renewal.

* The smoke treatment method, primarily used for creating black onyx, involves a two-step process where the stone is first impregnated with a sugar solution and then treated with acid to carbonize the sugar. This creates a permanent black color that penetrates the entire stone and is considered one of the most stable dyeing treatments available.

* Stabilization with dye involves treating stones with both colorants and hardening agents simultaneously. This technique is particularly important for softer or more porous stones like turquoise, where the treatment can improve both color and durability. The process can significantly enhance the stone's wearability in jewelry.

Modern gemstone dyeing represents a sophisticated blend of traditional techniques and contemporary technology. While each method has its advantages and limitations, the choice of treatment depends on various factors including the stone's natural characteristics, intended use, and desired outcome. Understanding these methods is crucial for proper disclosure in the trade and helps ensure appropriate care and maintenance of treated stones. The continued evolution of these techniques reflects the industry's ongoing effort to balance enhancement effectiveness with treatment stability and longevity.

How to Identify Dyed Gemstones



Being able to identify dyed gemstones is a crucial skill for both buyers and collectors. While professional gemmological testing provides the most definitive results, there are several visual and simple physical indicators that can help identify dyed stones. Understanding these signs allows consumers to make more informed decisions and helps ensure they pay appropriate prices for treated stones.

* Color concentration patterns are one of the most reliable indicators of dyeing. Natural stones typically show uniform or gradually changing colors, while dyed stones often display concentrated color in cracks, fissures, or along grain boundaries. This uneven distribution is particularly visible when examining the stone under magnification or strong light.

* The "bleeding test" involves gently rubbing the stone with a white cotton swab dampened with acetone or rubbing alcohol. If the stone is dyed, some color may transfer to the swab. However, this test should be performed cautiously as it can damage some stones and isn't conclusive for all types of dye treatments. Professional testing is always preferred for valuable pieces.

* Examination under ultraviolet light can reveal fluorescent reactions that differ from natural stones. Many dyes used in gemstone treatment fluoresce differently than their natural counterparts. For example, dyed jadeite often shows spotty fluorescence patterns that aren't present in natural material.

* Surface-level inspection using a 10x loupe or microscope often reveals concentrated color in surface-reaching fractures or around crystal boundaries. Natural stones typically don't show this type of color concentration pattern. Look particularly at areas where the stone meets its setting, as color concentrations are often visible there.

* The "hot point test," performed by professionals, involves touching a heated needle to an inconspicuous area of the stone. Dyed stones may show melting or color changes at the contact point, though this destructive test should only be performed by experienced gemmologists on appropriate materials.

* Unusual or unnatural colors for specific gem varieties often indicate dyeing. For instance, if you see perfectly matched strands of vibrant blue pearls or hot pink agate,

they're almost certainly dyed, as these colors don't occur naturally in these materials.

* Price comparison can be a practical indicator – if a stone's price seems unusually low compared to similar natural specimens, it's likely treated. For example, a strand of perfectly matched, intensely colored pearls selling for a fraction of the cost of natural pearls is almost certainly dyed.

* Angular color zoning or sharp color boundaries within a stone often indicate artificial treatment. Natural color zoning typically shows gradual transitions rather than sharp boundaries, though there are some exceptions in certain species.

Understanding how to identify dyed gemstones helps protect consumers and ensures fair market practices. While these identification methods can provide strong indications of treatment, definitive determination often requires laboratory testing by qualified gemmologists. The presence of dye treatment doesn't necessarily diminish a stone's beauty or value for fashion jewelry, but it should be reflected in the price and disclosed to all parties involved in the transaction. When in doubt about a stone's treatment status, consulting a qualified gemmologist can provide the most reliable assessment.

Advantages and Disadvantages of Dyed Gemstones

The use of dyed gemstones in jewelry presents both opportunities and challenges for

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the industry and consumers alike. Understanding these advantages and disadvantages is crucial for making informed purchasing decisions and maintaining realistic expectations about treated stones. While dyeing can make beautiful colors more accessible, it's important to consider all aspects of this enhancement method.

Pros of Dyed Gemstones

* The advantages of dyed gemstones extend beyond simple cost savings, offering unique opportunities for both jewellers and consumers in the modern market. These benefits have helped establish dyed stones as a significant segment of the jewelry industry, particularly in fashion jewelry and affordable luxury pieces, where they provide creative possibilities that might otherwise be unattainable.

* Market accessibility has dramatically increased through gemstone dyeing, allowing more consumers to enjoy colored stones. Natural stones of similar appearance might cost 5-10 times more, making dyed stones an important option for fashion jewelry and affordable luxury pieces.

* Color matching in dyed stones achieves consistency that would be nearly impossible or extremely expensive with natural stones. This is particularly valuable for matched strands of beads, multi-stone jewelry pieces, and sets where color uniformity enhances the overall design.

* Production costs remain significantly lower with dyed stones, making it possible to create attractive jewelry at more accessible price points. This cost efficiency extends throughout the entire production process, from raw material selection to final piece creation.

* Design flexibility increases dramatically with dyed stones, as jewellers can access specific colors consistently. This reliability allows for more creative designs and helps maintain quality standards across jewelry collections.

* Fashion trend adaptation becomes more feasible with dyed stones, as manufacturers can produce colors that match current style preferences. This responsiveness to market demands helps keep jewelry designs fresh and relevant.

The practical advantages of dyed gemstones have secured their place in modern jewelry making, offering valuable options for both industry professionals and consumers when properly disclosed and understood.

Cons of Dyed Gemstones

The limitations and drawbacks of dyed gemstones require careful consideration from both industry professionals and consumers. Understanding these challenges helps set appropriate expectations and ensures proper care and maintenance of treated stones, ultimately affecting their long-term value and wearability.

* Treatment durability remains a significant concern as many dyed stones can fade or change color over time. Exposure to sunlight, chemicals, heat, or regular wear may affect the stability of the treatment, potentially altering the stone's appearance.

* Value retention presents challenges as dyed stones typically depreciate more rapidly than natural ones. The treatment status often reduces both initial and resale value, making them less suitable for investment or heirloom pieces.

* Maintenance requirements are typically more demanding than natural stones. Dyed gems need protection from direct sunlight, harsh chemicals, heat exposure, and certain cleaning methods to maintain their appearance.

* Market limitations exist particularly in fine jewelry and investment markets, where natural stones are strongly preferred. This can affect both the stone's versatility and its future resale potential.

* Treatment disclosure requirements create additional responsibilities for sellers and sometimes lead to market confusion. Proper disclosure is both legally required and ethically necessary but can complicate sales processes.

The drawbacks of dyed gemstones primarily affect their long-term value and care requirements. While these limitations are significant, they don't necessarily diminish the stones' value for appropriate uses in fashion jewelry and affordable luxury pieces when properly understood and disclosed.

Ethics and Laws in Gemstone Dyeing

The ethical considerations surrounding dyed gemstones center primarily on transparency and proper disclosure. The jewelry industry's responsibility to inform consumers about treatments directly affects trust, value, and customer satisfaction. Understanding these ethical obligations helps maintain industry standards and protects both buyers and sellers in the gemstone market.


* Federal Trade Commission (FTC) guidelines explicitly require disclosure of all gemstone treatments that significantly affect value. This legal requirement mandates that sellers inform buyers about dyeing treatments at the point of sale, providing clear information about potential care requirements and value implications.

* Industry standards established by major gemmological laboratories and trade organizations require specific documentation of treatments. Professional certifications must note any color enhancement methods used, ensuring transparency throughout the supply chain.

* Value implications of dyeing treatments must be clearly communicated to customers. Sellers should explain how the treatment affects the stone's worth compared to natural alternatives, helping buyers make informed decisions based on their budget and preferences.


* Documentation requirements include maintaining records of treatment types and dates. Professional dealers should provide written documentation of any treatments, especially for significant purchases, helping protect both buyer and seller in future transactions.

* Customer education responsibilities extend beyond simple disclosure. Sellers should provide information about care requirements, durability expectations, and potential limitations of dyed stones to ensure customer satisfaction and proper maintenance.



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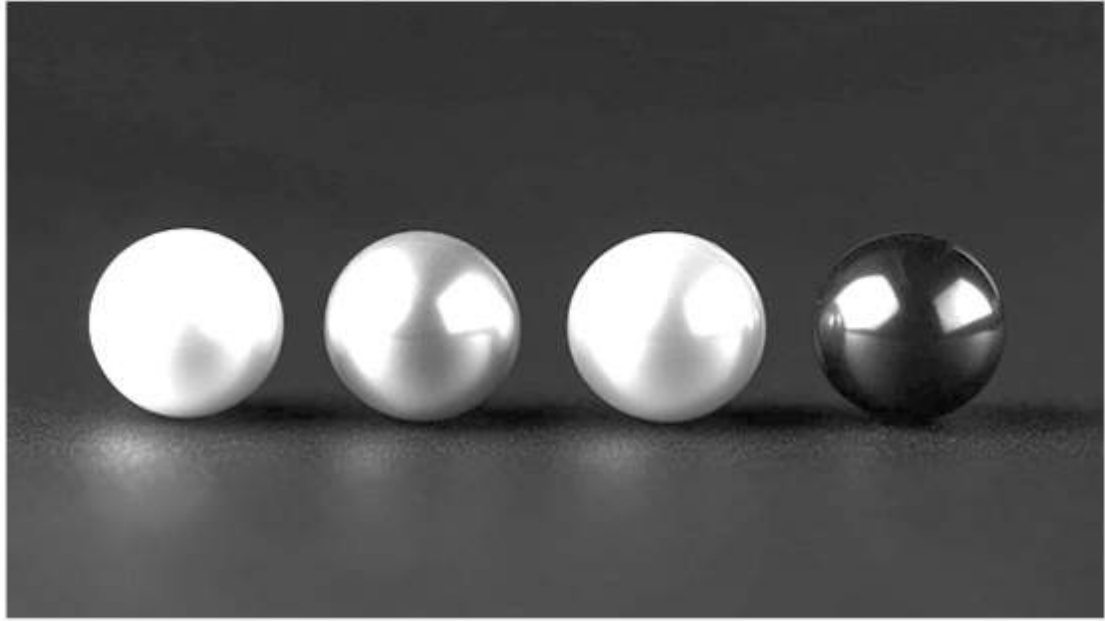
* Market transparency affects the entire supply chain, from miners to retail sellers. Each participant in the distribution chain has an obligation to maintain and pass along accurate information about treatments and enhancements.

* Professional ethics guidelines established by jewelry trade organizations specifically address treatment disclosure. These guidelines help standardize disclosure practices and maintain industry credibility.

* Consumer protection laws in many countries require explicit disclosure of treatments. Failure to disclose can result in legal penalties and damage to business reputation, emphasizing the importance of ethical business practices.

* Price structuring must reflect treatment status, with dyed stones appropriately priced below their natural counterparts. Ethical pricing practices help maintain market stability and consumer confidence.

The ethical framework surrounding dyed gemstones continues to evolve with changing technology and market demands. Proper disclosure and transparency remain fundamental to maintaining trust in the jewelry industry, protecting both businesses and consumers. When all participants understand and follow ethical guidelines, the market for dyed gemstones can function effectively as an important segment of the jewelry industry.



Proper care of dyed gemstones requires specific attention and understanding of their unique characteristics. Unlike natural stones, dyed gems can be particularly sensitive to environmental factors and common cleaning methods. Knowledge of proper care techniques helps maintain the stone's appearance and extends the longevity of the color treatment, ensuring the jewelry remains beautiful for years to come.

* Daily care requires gentle handling and protection from harsh environments. Remove dyed stone jewelry before swimming, bathing, or engaging in activities that involve chemicals or excessive heat. Even everyday products like perfume, lotion, or hairspray can potentially affect the dye's stability.

* Cleaning methods must be carefully chosen to avoid damaging the treatment. Use only lukewarm water and mild soap, gently cleaning with a soft cloth. Harsh detergents, ultrasonic cleaners, and steam cleaning can potentially damage or fade the dye treatment.

* Storage conditions play a crucial role in maintaining dyed stones. Keep pieces in individual soft cloth pouches or lined jewelry boxes to prevent scratching. Store away from direct sunlight and heat sources, as prolonged exposure can fade or alter the dye.

* Regular inspection helps identify potential issues early. Check for any color changes, fading, or bleeding, particularly in areas where the stone contacts skin or other materials. Early detection of problems allows for timely intervention.

* Professional cleaning should be done only by jewellers familiar with dyed stones. Always inform cleaning professionals about the treatment status of your stones to ensure appropriate cleaning methods are used.

* Environmental factors significantly impact dyed stones. Avoid exposing them to extreme temperatures, direct sunlight, or harsh chemicals. Even prolonged exposure to bright artificial light can potentially affect some dye treatments over time.

* Physical protection is essential as some dyed stones may be more fragile than their natural counterparts. Remove jewelry during activities that might subject stones to impact

or abrasion. This includes household chores, sports, or other physical activities.

* Temperature changes can affect dyed stones more severely than natural ones. Avoid exposing jewelry to sudden temperature changes, which can potentially stress the stone and affect the dye stability.

* Maintenance schedules should include regular gentle cleaning and professional inspection at least annually. This helps ensure any potential issues are identified and addressed before they become serious problems.

Proper care of dyed gemstones extends beyond simple cleaning and storage. It requires an understanding of the treatment's nature and potential vulnerabilities. While these care requirements might seem demanding, they're essential for maintaining the beauty and longevity of dyed stone jewelry. When properly cared for, dyed gemstones can provide years of enjoyment, making them a valuable option for fashion jewelry and everyday wear.



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INKJET PRINT HEAD



DIN-10358157

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(South Gujarat University-PhD Notification no : 394 dated April 21, 1994)

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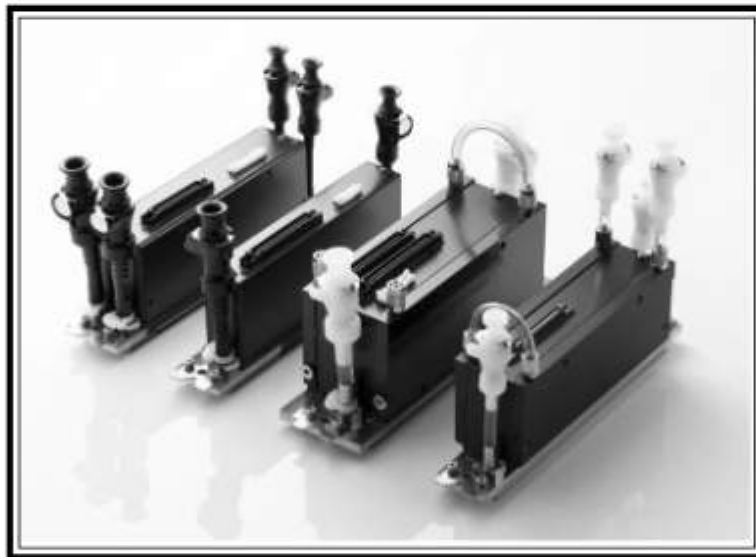
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Inkjet printers use various print head technologies to deliver ink onto paper.

1. Thermal Inkjet (Bubble Jet):

Description : Uses heat to vaporize ink, creating a bubble that forces ink droplets out of the Nozzle.

Mechanism : A heating element rapidly heats the ink, forming a bubble that expands and ejects the ink.

Advantages : Cost-effective, compact, widely used in consumer printers.

Disadvantages : Limited lifespan due to heat stress, potential for nozzle clogging.

Applications : Home and office printers, affordable desktop models.

2. Piezoelectric Inkjet :

Description : Uses a piezoelectric crystal that vibrates when an electric current is applied, pushing ink out of the nozzle.

Mechanism : The crystal deforms under voltage, creating pressure to eject ink droplets.

Advantages : Precise control over droplet size, supports a wider range of inks, longer lifespan.

Disadvantages : More expensive to manufacture, complex design.

Applications : High-end printers, industrial printing, wide-format printers, Labels.

3. Continuous Inkjet (CIJ):

Description : A continuous stream of ink droplets is generated, with only selected droplets directed to the paper, others are recycled.

Mechanism : Ink is pumped through a nozzle, and electrostatic fields control droplet placement.

Advantages : High-speed printing, suitable for rough surfaces, low maintenance.

Disadvantages : Complex system, higher ink waste, less common in consumer printers.

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Applications : Industrial applications like packaging, coding and marking.

4. Drop-on-Demand (DoD) :

Description : Ejects ink droplets only when needed, unlike continuous inkjet. Includes both thermal and piezoelectric methods.

Mechanism : Ink is expelled only for specific print demands, reducing waste.

Advantages : Efficient ink use, high precision, versatile.

Disadvantages: Slower than CIJ for some applications.

Applications : Most consumer and professional inkjet printers.

5. Electrostatic Inkjet :

Description : Uses electrostatic forces to control ink droplet ejection, less common in modern printers.

Mechanism : Charged ink droplets are deflected by electric fields onto the paper.

Advantages : High precision for specific applications.

Disadvantages : Complex and costly, limited use in mainstream printing.

Applications : Specialized printing, such as high-resolution graphics.

6. Microfluidic Inkjet :

Description : Utilizes microfluidic channels to control ink flow with high precision, often combined with thermal or piezoelectric mechanisms.

Mechanism : Microfluidic systems manipulate tiny volumes of ink through precisely engineered channels, enabling accurate droplet formation.

Advantages : Exceptional precision, reduced ink waste, suitable for advanced applications like 3D printing or bio-printing.

Disadvantages : Complex and expensive to produce, still emerging in commercial printers.



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TOOTHPASTE

INFORMATION HOW IT IS MANUFACTURED



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Toothpaste is a paste or gel used with a toothbrush to clean and maintain the appearance of teeth. It typically contains a combination of ingredients that help remove plaque, prevent tooth decay, and freshen breath.



Technical Details

1. pH Level: Toothpaste typically has a pH level close to neutral (pH 7) to ensure it doesn't harm teeth or gums.

-
-
2. Viscosity: Toothpaste has a specific viscosity that allows it to flow easily off the toothbrush and stay on the teeth during brushing.
 3. Raw Materials
 4. Abrasives: Silica, calcium carbonate, or baking soda help remove plaque and surface stains.
 5. Detergents: Surfactants like sodium lauryl sulfate create a rich lather and help clean teeth.
 6. Fluoride: Sodium fluoride or stannous fluoride help prevent tooth decay and strengthen tooth enamel.
 7. Humectants: Glycerine or sorbitol help retain moisture and prevent toothpaste from drying out.
 8. Flavouring Agents: Natural or artificial flavourings give toothpaste its desired taste.
 9. Thickeners: Carrageenan or xanthan gum help maintain the toothpaste's texture and consistency.

Manufacturing Process

1. Mixing: Raw materials are mixed together in a specific order and ratio to create a uniform blend.
2. Milling: The mixture is then milled to ensure the desired particle size and texture.
3. Blending: The milled mixture is blended with other ingredients, such as flavourings and humectants.
4. Filling: The toothpaste is then filled into tubes or containers.
5. Packaging: The filled tubes or containers are labelled and packaged for distribution.



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HOW SPECIALTY INKS LIKE FLUORESCENTS & METALLICS ARE TRANSFORMING VISUAL COMMUNICATION IN SCREEN PRINTING



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In today's oversaturated visual world, the demand for standout graphics is higher than ever. Whether it's for branding, fashion, or safety, ordinary inks often fail to leave a lasting impression. That's where fluorescent and metallic inks come in, offering vividness and brilliance that capture attention instantly.

1. What Are Fluorescent and Metallic Inks?

Fluorescent Inks are highly pigmented inks that emit intense color under normal lighting and glow under UV light. They are perfect for visibility-focused designs.

Metallic Inks contain tiny reflective particles like aluminum or bronze, creating a shiny, mirror-like finish that simulates gold, silver, or other metallic tones.

2. Visual Impact

These specialty inks are not just about color - they're about experience. Fluorescent inks evoke energy and dynamism, often used in party merchandise, athletic wear, and promotions. Metallic inks exude luxury and prestige, perfect for wedding cards, cosmetic packaging, and premium branding.

Their tactile and visual properties help differentiate brands and products in a crowded marketplace.



3. Technical Considerations in Screen Printing

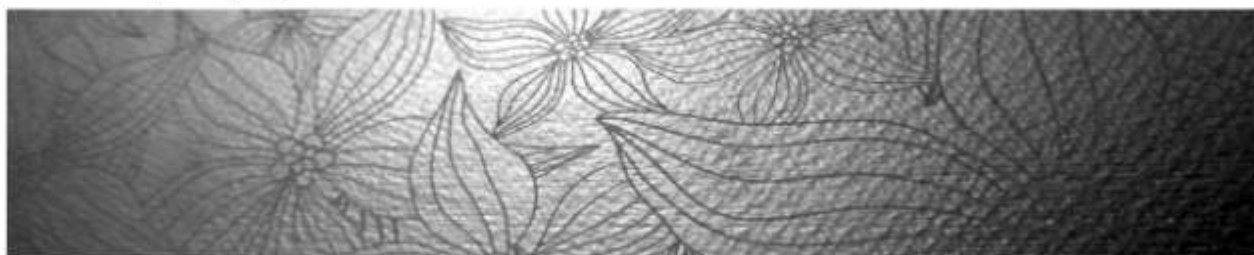
Both types of inks require specific mesh counts and stencil setups for optimal results.

Fluorescent inks are best applied on light backgrounds and may require a white under base for opacity.

Metallic inks perform well with coarser meshes and benefit from slower printing speeds for proper flake alignment.

4. Application Versatility

Fluorescent Applications: Safety vests, event merchandise, children's books, promotional posters. Metallic Applications: Greeting cards, electronics branding, fashion apparel, decorative packaging.



Fluorescent and metallic inks are more than aesthetic tools—they are strategic assets in print communication. By incorporating these into your printing arsenal, you bring life, light, and luxury to every design.



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NANOFILTRATION (NF) VS REVERSE OSMOSIS (RO)



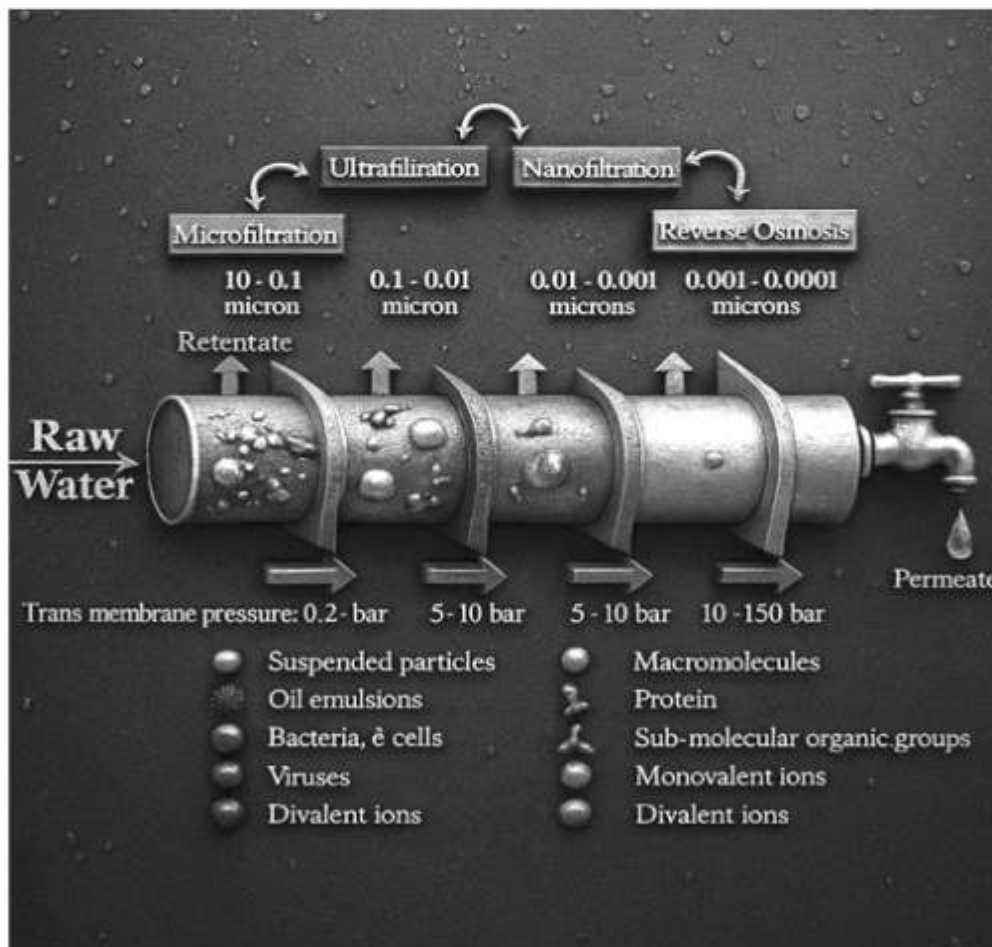
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Nanofiltration (NF)

Features :

- * Pore size ranges from 0.01 to 0.001 microns.
- * Operates at moderate pressure (4–30 bar).
- * Removes bacteria, viruses, organics, and divalent ions (like Ca^{2+} , Mg^{2+}).
- * Does not effectively remove monovalent ions like Na^+ and Cl^- .
- * Retains some minerals, which makes the water taste more natural.
- * Energy consumption is lower than RO.
- * Higher water recovery (60–85%) compared to RO.
- * Less prone to membrane fouling.
- * Suitable for softening, dye removal, and partial demineralization.
- * Commonly used in food processing, water softening, and as a pre-treatment for RO.

Reverse Osmosis (RO)

Features:

- * Pore size is much smaller (~ 0.001 to 0.0001 microns).
- * Requires high pressure (10–70 bar or more).
- * Removes nearly all ions (both monovalent and divalent), bacteria, viruses, and organics.
- * Produces ultrapure water with no taste due to mineral removal.
- * Higher energy consumption than NF.
- * Moderate water recovery (typically 35–75%).

-
-
- * More prone to membrane fouling if not properly pre-treated.
 - * Ideal for desalination of seawater or brackish water.
 - * Used in pharmaceuticals, electronics, and drinking water systems.
 - * Requires more robust infrastructure and maintenance.

Which one to choose & why?

1. Choose NF if:

- * You want to remove hardness and organics but keep some minerals.
- * Energy savings and lower operating costs are important.
- * You're treating surface water, well water, or industrial wastewater.
- * Taste and natural mineral content are preferred in the final water.

2. Choose RO if:

- * You need maximum purity, such as for desalination or pharma applications.
- * You must remove all dissolved salts, including monovalent ions like sodium.
- * Your water source is salty or brackish.
- * You can handle higher costs and energy usages.

In summary:

- * Use NF for cost-effective, mineral-retaining filtration.
- * Use RO for complete purification and salt removal.



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SULFUR DYES

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(Article found during literature Survey)

I. INTRODUCTION

Historically, the name "sulfur dyes" has been applied to both a chemical class of dyes and a dyeing class. The first sulfur dyes were all made by thionation of various organic materials with either sulfur or sodium polysulfide, and they were all applied to cotton from sodium sulfide solution. It was not long before other types of dye were made by thionation procedures, in particular the Hydron Blue type of vat dye. More recently, wholly synthetic dyes have appeared on the market which are applied from sulfide solution. While these are sulfur dyes from the point of view of the dyer, to the chemist they belong in the azo, phthalo-cyanine, anthraquinone, or oxazine class. It seems reasonable to define as sulfur dyes those dyes that are best applied from a sulfide dyebath, and this definition is adopted here.

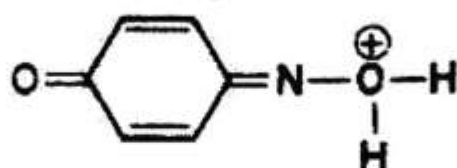
Although cotton and other cellulosic fibers form a declining proportion of world fiber production, they still constitute a very large and important market and are likely to do so in the foreseeable future. Sulfur dyes are widely used for the production of fast dyeings on all kinds of cellulosic materials, especially in heavy shades where their cost advantage over vat dyes is greatest. During the last 20 years, there have been many changes in the sulfur dye market. Cotton raw -stock dyeing is now much less common, but there has been a corresponding increase in continuous dyeing of piece goods and batch dyeing of yarn both in package and beam form. A number of dye manufacturers have ceased to offer sulfur dyes. In England there is now only one manufacturer, and in the United States two manufacturers have gone out of the business since 1965. Nevertheless, the sulfur dye market still continues to grow now and is likely to do so while the manufacturers continue

to improve their product line to meet changing circumstances.

This review covers progress in the chemistry, manufacture, and application of sulfur dyes since 1950, and is intended to supplement the chapter on sulfur dyes in CSD II. The reader is assumed to be familiar with that work.

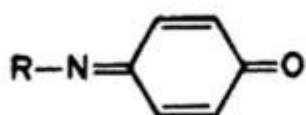
II. INTERMEDIATES

The condensation of p-nitrosophenol with amines in sulfuric acid, giving indophenols, has been studied. It is concluded that in 75% sulfuric acid nitrosophenol reacts in the protonated quinonoid form (I).

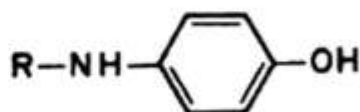


(I)

This seems reasonable in view of the fact that the products are the quinonoid indophenols (II), rather than the leucoindo-phenols (III).



(II)



(III)

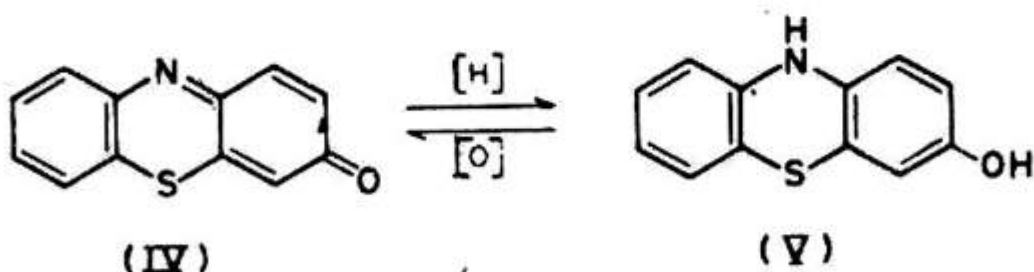
Kinetic studies of the formation of o-toluidine indophenol show that the reaction is of second order with respect to both o-toluidine and p-nitroso-phenol, and the reaction rate diminishes with sulfuric acid concentration. These results are interpreted as indicating reaction between the free base of o-toluidine and the conjugate acid of p-nitroso-phenol. Similar results were obtained for the formation of di-methyl-aniline and diphenylamine indophenols. The rate constant for the reaction with diphenylamine was shown to be several

hundred times greater than for reaction with o-toluidine and dimethylaniline. This is in accordance with industrial practice, since diphenylamine indophenol may be made at -10°, while o-toluidine and dimethylaniline require temperatures between 6° and 10°. It has been shown that the troublesome deterioration of o-toluidine indophenol at higher temperatures is due to polymerization.

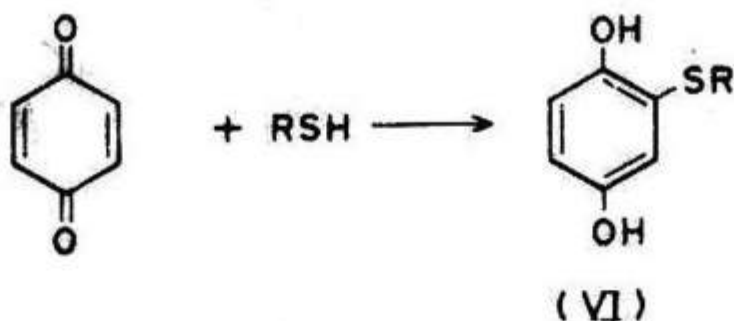
III. The Chemistry of Thionation

Conventional sulfur dyes are still, for the main part, of unknown and variable constitution. Reproduction of shade from batch to batch is achieved partially by arbitrary control of reaction conditions and partially by checking the course of the thionation by means of laboratory dyeings. The lack of reliable structural formulas for these dyes hinders scientific control of industrial processes and discourages the academic research worker from any investigation of reaction mechanism.

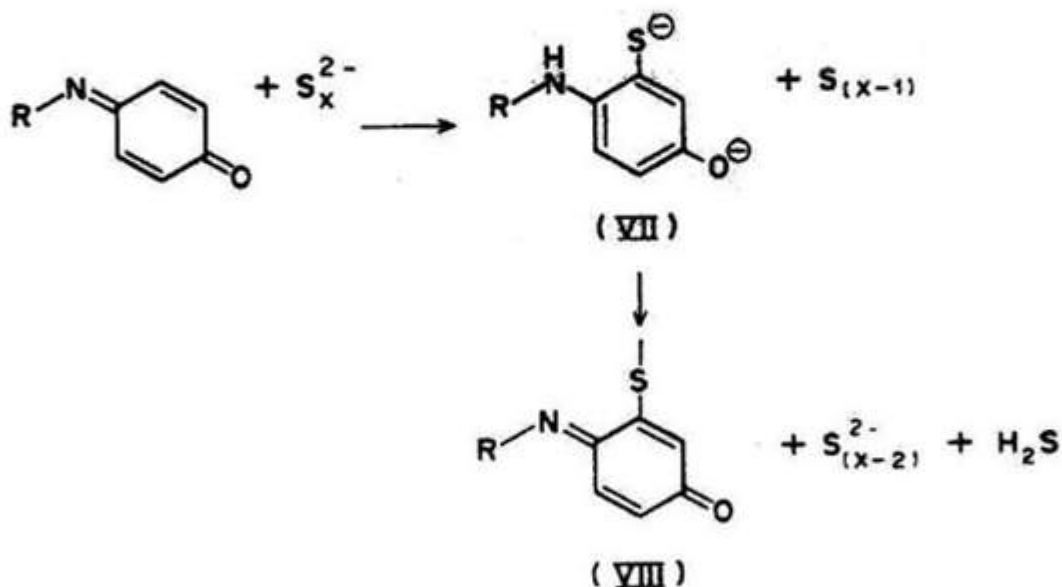
However, certain classes of dye have received much attention over the years, and despite conflicting evidence there is a measure of agreement on certain structures. In particular, the indophenol dyes are fairly well understood. This important class of dyes includes the blues and greens, the Indocarbon Blacks, and the red-browns derived from 4-hydroxydiphenylamine. It is agreed that the fundamental chromophore is the phenothiazinone group (IV), which accounts for the high color value of these dyes and




for their reduction to pale yellow leuco compounds (V). It is further agreed that as many as three more sulfur atoms are attached to the quinoneimine ring. The formation of such structures from indophenol may be explained in terms of nucleophilic substitution by polysulfide ions. It is known that quinones react readily with mercaptans, forming substituted quinols (VI). The analogous reaction of polysulfide with quinone-imine is



postulated as the first stage in a thionation, giving the substituted leuco-indophenol (VII). This is then dehydrogenated by sulfur to the quinone-imine form (VIII), liberating hydrogen sulfide. This theory accounts for the inability of sodium sulfide to form sulfur dyes with indophenols, since sulfide is unable to oxidize the leuco form of the intermediates to the corresponding quinone-imine. This process of substitution followed by oxidation is repeated until thionation is complete and the dye is formed.

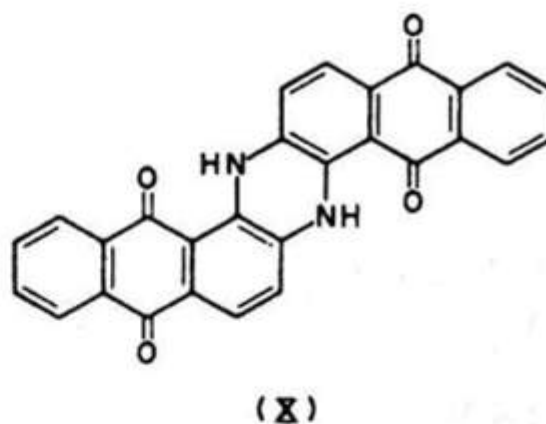
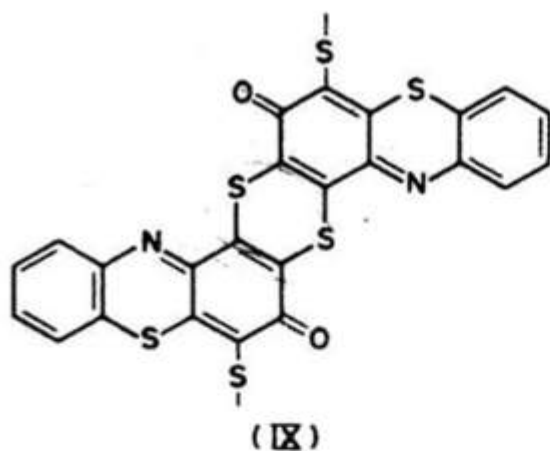


The work of Zerweck, Ritter, and Schubert indicates the presence of a thianthrene structure (IX) in indophenol dyes. This structure might reasonably be expected to confer

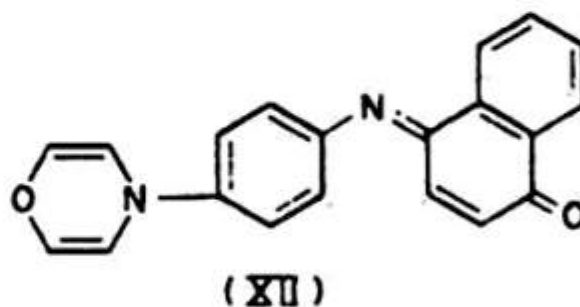
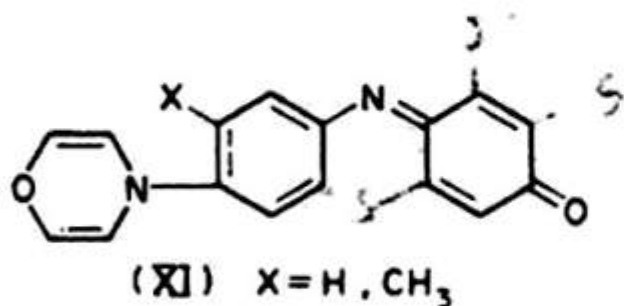


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affinity for cotton, since it is isosteric with a number of vat dyes, e.g., indanthrone (X). Support for the thianthrene ring system is found in the properties of a group of sulfur dyes made from N-arylmorpholines. The dyes derived from the indophenols (XI) were blue and had affinity for cotton, whereas the dye from the naphthoquinone-imine

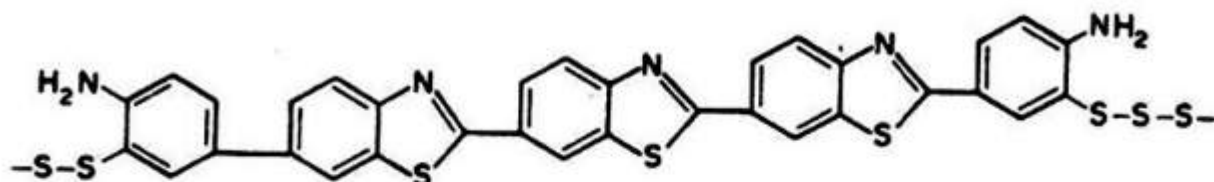


indophenol (XII) was violet and had no affinity for cotton. The lack of affinity and violet color of this dye may be accounted for by the inability of the naphthoquinone-imine to form a thianthrene bridge with a second molecule.

Sulfur Black T has been the subject of various investigations. Hiayama has studied the positions into which sulfur enters to form the dye. Analysis of sulfur black free from uncombined sulfur shows that a minimum of five sulfur atoms is needed for each 24 carbon atoms in order to form the dye. The effect of polysulfide composition on yield has been studied. It is claimed that the maximum yield is obtained with Na_2S_4 . Other authors contend that the optimum polysulfide is between $Na_2S_{3.9}$ and Na_2S_5 , depending on the molar ratio of poly-sulfide to dinitrophenol. These results are in agreement with

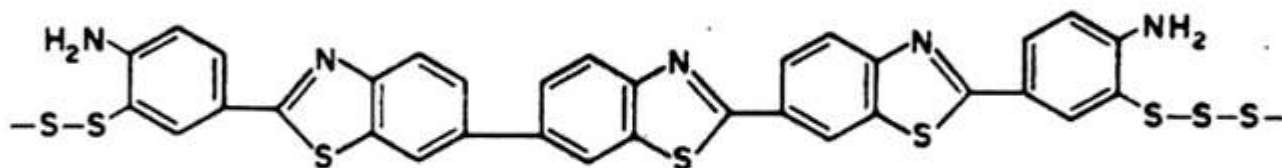
industrial thionation procedures, which utilize $\text{Na}_2\text{S}_{3.6}$ to $\text{Na}_2\text{S}_{4.3}$.

A degradative study of Immedial Supra Yellow GWL indicates that it consists mainly of (XIII). The dye is therefore made by a sulfur bake of benzidine with either Primuline



(XIII)

Base, p-toluidine, or dehydrothio-p-toluidine and p-toluidine. Most probably Primuline Base is used, as the other alternatives would be expected to give rise to such structures as (XIV).



(XIV)

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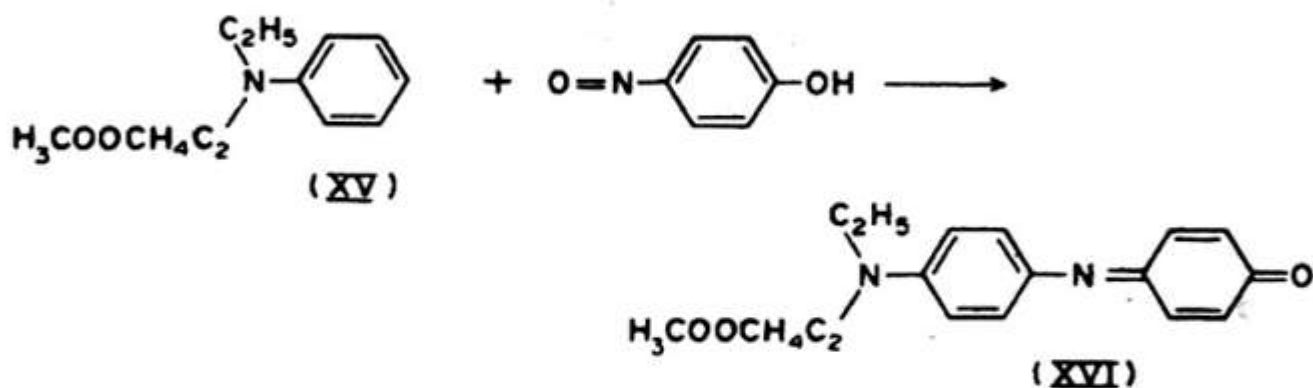
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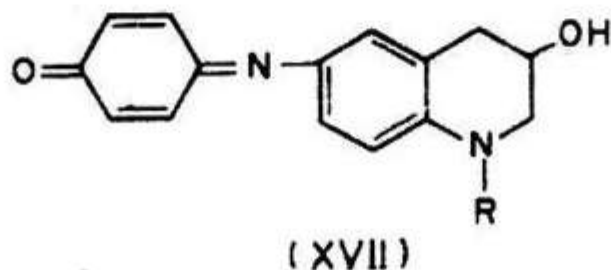
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A. INDOPHENOLS

There has been some interest in Japan in sulfur dyes for Vinyon. Thionation of indophenols carrying carbo-methoxyalkyl groups gives blue dyes. The intermediates are made by condensation of anilines with methyl acrylate. For instance, N-ethylaniline is heated with methyl acrylate and a catalytic quantity of hydroquinone in acetic acid. The resulting amine (XV) is condensed with p-nitroso-phenol to form the corresponding indophenol (XVI). Indophenols having hydroxyethyl groups are claimed to be applicable to synthetic fibers.



Some of these indophenols are made by condensation of p-nitro-sophenol with much amines as N,N-2-hydroxyethylallylaniline. Others are made by oxidative condensation of 1-naphthol with the appropriate p-amino-NN-dialkylaniline. Similarly, N-arylmorpholines yield indophenols which are thionated to give violet and blue dyes for acrylic fibers. Condensation of p-aminophenol with tetrahydro-1-alkyl-3-quinolinols in the presence of an oxidant gives indophenols (XVII) which are thionated in poly-sulfide to yield blue dyes of good lightfastness for Vinyon. Other blue and green dyes for Vinyon are made



by thionating the indophenols obtained by oxidative condensation of N,N-dialkyl-phenylene-diamines with cresols and xylenols.

A number of blue indophenol sulfur dyes have been made from N-alkyldiphenylamines, Methyl, ethyl, and butyl di-phenylamines yield bright greenish blue dyes, the methyl derivative being notably fast to chlorine. N-Alkyl-3-chlorodiphenylamines give significantly redder shades, but chlorine in the 4-position has little effect on shade.

The blue dyes made from 2,4-dinitro-4'-hydroxydiphenylamine are much improved in shade by a treatment with sodium cyanide in conjunction with an alkaline oxidation, the shade becoming significantly redder and brighter. A similar cyanide treatment is claimed to improve the exhaust dyeing properties of Sulfur Black.

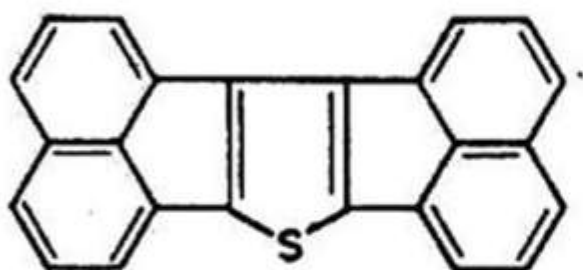
B. POLYCYCLIC INTERMEDIATES

The commercial success of the lightfast orange dyes made by a sulfur bake of decacyclene has led to some activity in this and related areas.

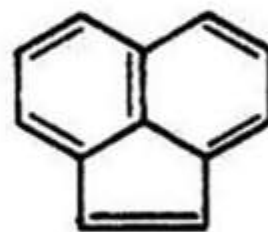
Chlorination of decacyclene with sulfuryl chloride at 70° using an iron catalyst introduces 10 atoms of chlorine. This intermediate is baked with sulfur at 250°-270° for 6-8 hours and then desulfurized to yield a deep brown sulfur dye of good lightfastness. The various nitro-decacyclenes are thionated with sulfur chloride in chloro-sulfonic acid to yield brown sulfur dyes of outstanding fastness to light. Tri-nitro-decacyclene and sulfur are stirred in chloro-sulfonic acid at 20°. Chlorine is introduced, allowing the temperature to rise to 50°, and the mixture is stirred for 3 hours. It is then drowned in ice, and the precipitated dye is filtered off, washed, and desulfurized to give a yellow-brown dye of high light-fastness. Similarly, hexanitrodecacyclene yields a khaki dye, and enneanitrodecacyclene a black-brown dye, Decacyclene is brominated and then thionated in sulfur to give orange and yellow-brown sulfur dyes of excellent light-and wash-fastness. For example, bromination of decacyclene at room temperature with an iodine catalyst in chloro-benzene gives a tribromodecacyclene. This is thionated by baking with sulfur at 195-200 to give yellow-orange shades. Thionation at 250° gives red-brown shades. It is interesting that the brominated decacyclene can be thionated at such a relatively low

temperature, whereas deca-cyclene itself requires temperatures in excess of 300°. The brominated intermediate evolves hydrogen bromide and hydrogen sulfide during thionation, so both bromine and hydrogen atoms are displaced by sulfur.

Decacyclene is itself made by a sulfur bake process in which acenaph-thene is dehydrogenated with evolution of hydrogen sulfide. It would be economically very attractive if cheap acenaphthene could be directly converted to decacyclene orange by sulfur fusion. However, this has not been achieved. A considerable quantity of diacenaphtho[1,2-b:1,2'-d]-thiophene (XVIII) is produced as a by-product, and further thionation of this yields brown dyes of inferior lightfastness. Similar red-brown shades are obtained by sulfur bakes of acenaphthylene (XIX). However, it is claimed that the dyes made by polysulfide bakes of nitro-acenaphthenes have superior lightfastness. 5-Nitroacenaphthene is baked with Na_2S_8 at 240°-260° to give a dye which gives red-brown shades from a violet-brown vat. Pure



(XVIII)

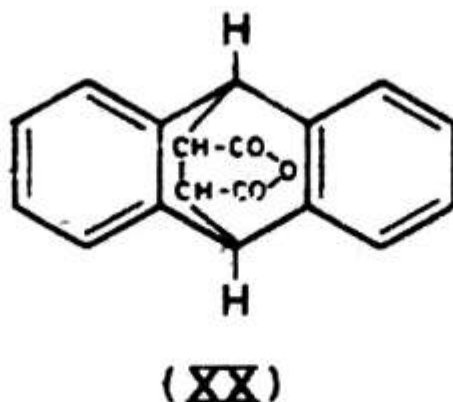


(XIX)

5,6-dinitroacenaphthene gives black-brown shades, and the addition of other dinitroacenaphthene isomers gives this dye a redder tone. These products are claimed to be suitable for cotton, nylon, and Vinylon.

Sulfur bakes of pyrene, polychloropyrene, and acenaphthene in the presence of copper salts are claimed to give red-brown dyes of excellent light-and wash-fastness. Pyrene is baked with sulfur at 280° - 285° for 4 hours. The dye is isolated by dissolving the crude melt in caustic soda solution and acidifying with hydrochloric acid. Methylpyrenes have been shown to yield valuable sulfur dyes. 3-Methylpyrene is baked with sulfur at 240°-260° to give a brown sulfur dye. The 4-methyl isomer gives a redder shade of brown. Addition of p-phenylenediamine to this melt gives yellow-olive to yellow-brown yellow-

brown shades. Similarly, 4-methyl-pyrene with o-phenylenediamine gives a brown dye, with p-toluidine a red-brown dye, and with 2-aminoanthraquinone a red-brown dye. Condensation of anthracene with maleic anhydride at 250°-260° gives 9,10-dihydroanthracene-9,10-endo-a, b-succinic anhydride (XX). When this intermediate is baked

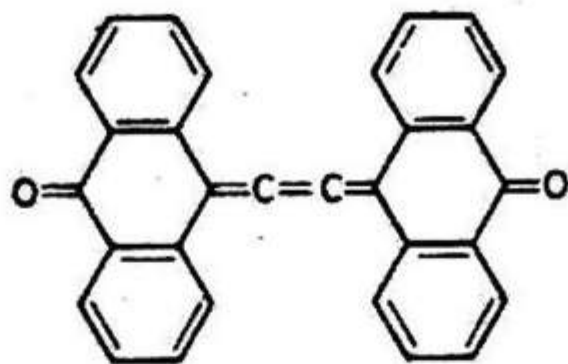


with sulfur the product is insoluble in sulfide, but dyes cotton dark olive from a hydrosulfite vat. However, brown sulfur dyes are obtained if 2,4-diaminotoluene, benzidine, or dehydrothio-p-toluidine is added to the thionation charge.

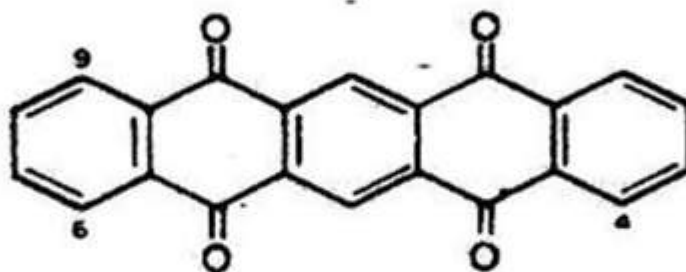
C. ANTHRAQUINONE DERIVATIVES

Thionation of 2-aminoanthraquinone and its halogen and sulfonic acid derivatives with sodium polysulfide gives olive and brown vat dyes. Thus 2-aminoanthraquinone is baked with Na₂S_{5.7} at 280° for 12 hours, giving a vat dye which dyes cotton in black-brown shades, from a dark brown vat. Sodium 2-aminoanthraquinone-3-sulfonate baked with Na₂S₅ at 280°-300° for 12 hours gives an olive-brown vat dye. A similar dye is obtained from 2-amino-3-chloroanthraquinone, while 2-amino-1-chloroanthraquinone gives an olive-gray and sodium 2-amino-1-bromoanthraquinone-3-sulfonate, an olive-brown. These dyes have excellent fastness to washing and light, but not to chlorine.

Sulfur dyes are obtained by polysulfide bakes of various 1-hydroxy-anthraquinone derivatives. Reflux thionation of di-anthronethylene (XXI) in ethanolic sodium polysulfide gives a red-violet product which dyes cotton pink from a red hydrosulfite vat. Under similar conditions, 2,2'-dichlorodianthron-ethylene yields a violet vat dye. Polysulfide thionations of pentacene-di-quinones (XXII) and their precursors give rise to fast brown and



(XXI)



(XXII)

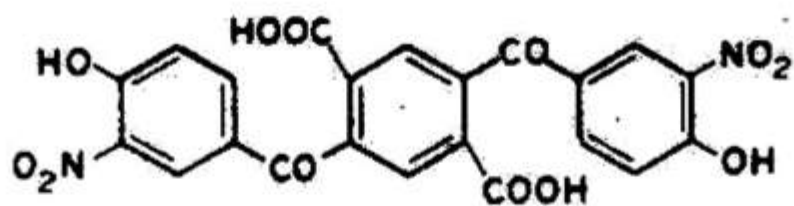
olive-green dyes. Thus, baking 1,6-diacetylaminopentacenediquinone at 270° with sodium tetrasulfide forms an olive dye, while the 1,4,6,9-tetraphenoxy and the 1-amino derivatives give olive-green shades. The substituted di-benzoyl-benzene-dicarboxylic acids (XXIII) and (XXIV) yield an olive-brown dye. Condensation of pyromellitic dianhydride with chlorobenzene under Friedel-Crafts conditions gives 4,6-di-p-chlorobenzoylisophthalic acid (XXV). Nitration, alkaline hydrolysis, and sulfide reduction leads to the diamine (XXVI). Condensation with chloranil gives the dioxazine (XXVII), which may be thionated to yield an olive dye.

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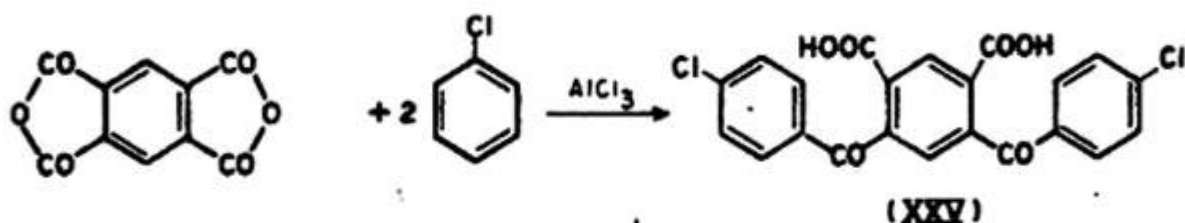


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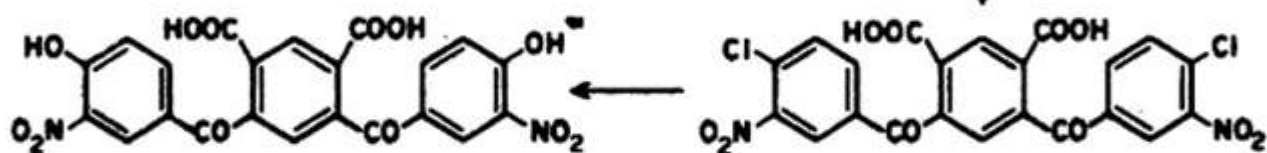




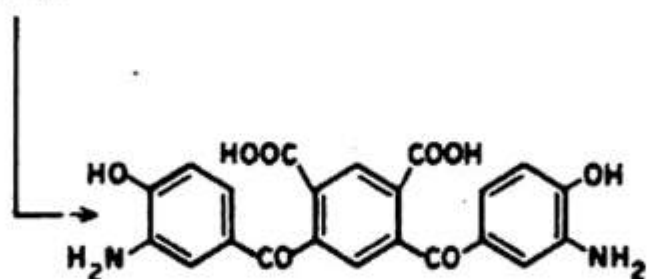
(XXIV)



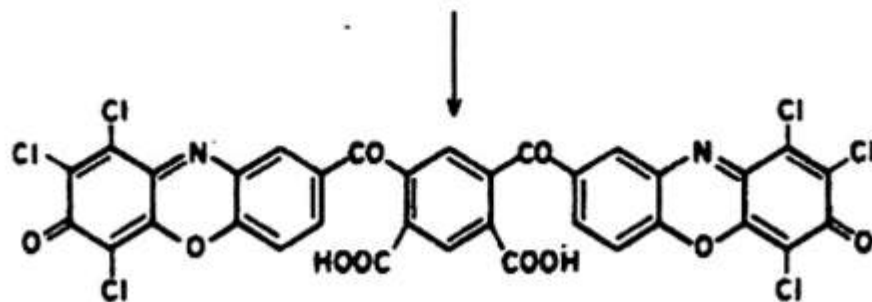
(XXV)



(XXIII)



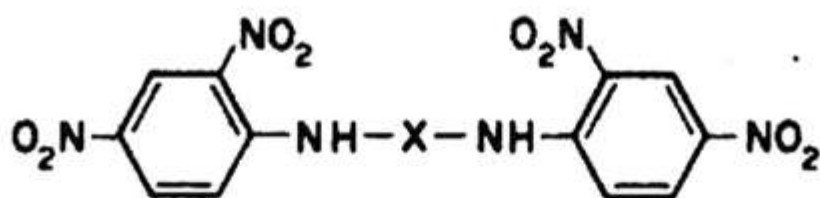
(XXVI)



(XXVII)

D. AROMATIC AMINES

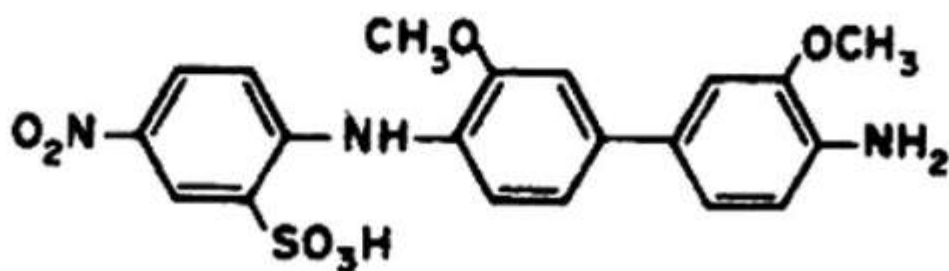
Brown sulfur dyes are made by baking compounds of general formula (XXVIII) with polysulfide. Examples of such intermediates are N,N'-bis(2,4-dinitrophenyl)ethylenediamine



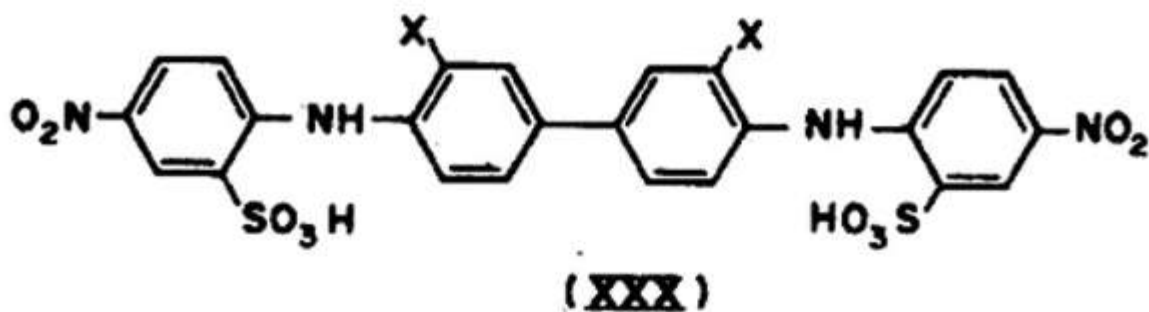
(XXVIII)

and N,N'-bis[2-(2,4-dinitrophenylamino) ethyl]amine, made by condensation of 2 moles of 2,4-dinitrochlorobenzene with the appropriate diamine.

Diphenylamine derivatives useful as sulfur dye intermediates are made by condensation of o-tolidine and o-di-anisidine with 1 or 2 moles of a chloro-nitrobenzene-sulfonic acid. Thus o-di-anisidine is condensed with 2-chloro-5-nitrobenzenesulfonic acid to give (XXIX). This is baked with polysulfide to give brown shades. The analogous intermediate from o-tolidine gave a green-gray sulfur dye. The intermediates (XXX) made from 2 moles of the reactive chloro compound, give gray dyes. It is claimed that ethylbenzene derivatives yield sulfur dyes superior to those derived from the homologous toluene derivatives. In particular, dinitro-, diamino-, and amino-nitro-ethylbenzenes give useful olive-green and olive shades when baked with polysulfide.



(XXIX).



E. COPPER PHTHALOCYANINE

Conventional sulfur and polysulfide bakes of copper phthalocyanine do not yield any sulfur dye. This intermediate can be thionated with the aluminum chloride/sulfur chloride complex, but the dark green sulfur dyes so made have rather poor solubility and have not been marketed. Brighter shades of green with better solubility are made by chloro-sulfonation of copper phthalocyanine, followed by zinc or iron reduction of the sulfonyl chloride to mercaptan. Thus copper phthalocyanine is heated with chloro-sulfonic acid for 3 hours at 145°-150°, and the mixture is then poured onto ice. Zinc dust is added and the mixture is heated to 90° to complete the reduction. However, if the chloro-sulfonation temperature is kept down to 100°-110°, iron reduction yields blue sulfur dyes. Presumably fewer mercapto groups are introduced than in the green dyes. These products may be further thionated by baking with sulfur to give dark green sulfur dyes of higher sulfur content. Thus copper phthalocyanine is chloro-sulfonated at 145° and the product is reduced with iron and acid. The resulting green powder is baked with sulfur at temperatures between 220° and 245° to give strong dark green sulfur dyes of high light and wash-fastness, the lower temperatures giving a yellower shade. Similar results are obtained if the mercapto-phthalocyanines are methylated before being thionated. Similarly tetra-(4)-thiocyanatocopper phthalocyanine is baked with sulfur at 240-265° to give dark green sulfur dyes.

Addition of hydroxylamine salts and a catalytic quantity of molybdenum or vanadium oxide to the chloro-sulfonation has a drastic effect on the shade of the dye. Iron reduction of the chloro-sulfonation product gives green-gray sulfur dyes of good lightfastness. Copper phthalocyanine with a trace of molybdenum oxide is added to an excess of chloro-sulfonic acid containing 8 moles of hydroxylamine sulfate. The mixture is heated at 150° for 2

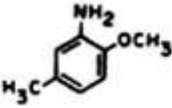
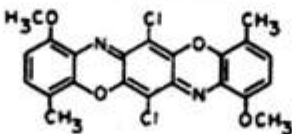
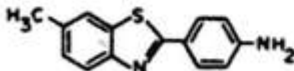
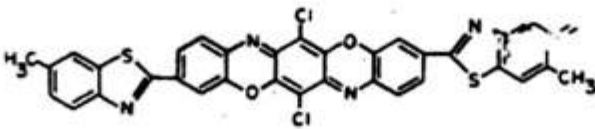
hours, added to ice and an excess of iron powder, and then stirred at 50 for 24 hours. The product is filtered off and washed acid-free. Addition of sulfur chloride to the chloro-sulfonation changes the shade of the final dye to violet-gray. Thus, copper phthalocyanine is added to 3 moles of hydroxylamine sulfate and a trace of ammonium molybdate in excess chloro-sulfonic acid. The mixture is heated to 125° and sulfur chloride (20 moles) is added dropwise. After 1-2 hours further heating at 125° the mixture is drowned in ice, iron powder, and sulfuric acid. The reduction is completed as above. A neutral gray shade is obtained if 50 to 90% of the chloro-sulfonic acid is replaced by 100% sulfuric acid in the above procedure. If the iron reduction stage is omitted the product is a greenish-gray sulfur dye. The exhaust dyeing properties are improved by keeping the reaction temperature below 100°.

Copper phthalocyanine sulfonyl chlorides may be reduced with sodium sulfide or sodium hydrosulfide to give copper phthalocyanine sodium sulfinates, which may also have sulfonate or thiol-sulfonate groups. These are heated in an autoclave with sodium thiosulfate or sodium polysulfide to give green sulfur dyes. Alternatively the sulfonyl chloride may be treated with thiourea to yield bright green sulfur dyes.

F. DIOXAZINES

Dioxazine intermediates, like phthalocyanines, are not readily thionated by conventional procedures, but yield sulfur dyes by iron sulfur dyes of varying shades. The starting amine, dioxazine intermediate, and the shade of dye obtained are listed in Table I.

TABLE I
THIONATION OF DYES FROM DIOXAZINES

<i>Amine</i>	<i>Dioxazine</i>	<i>Color of vat</i>	<i>Shade on cotton</i>
		Green-blue	Red-violet
		Yellow	Blue-violet

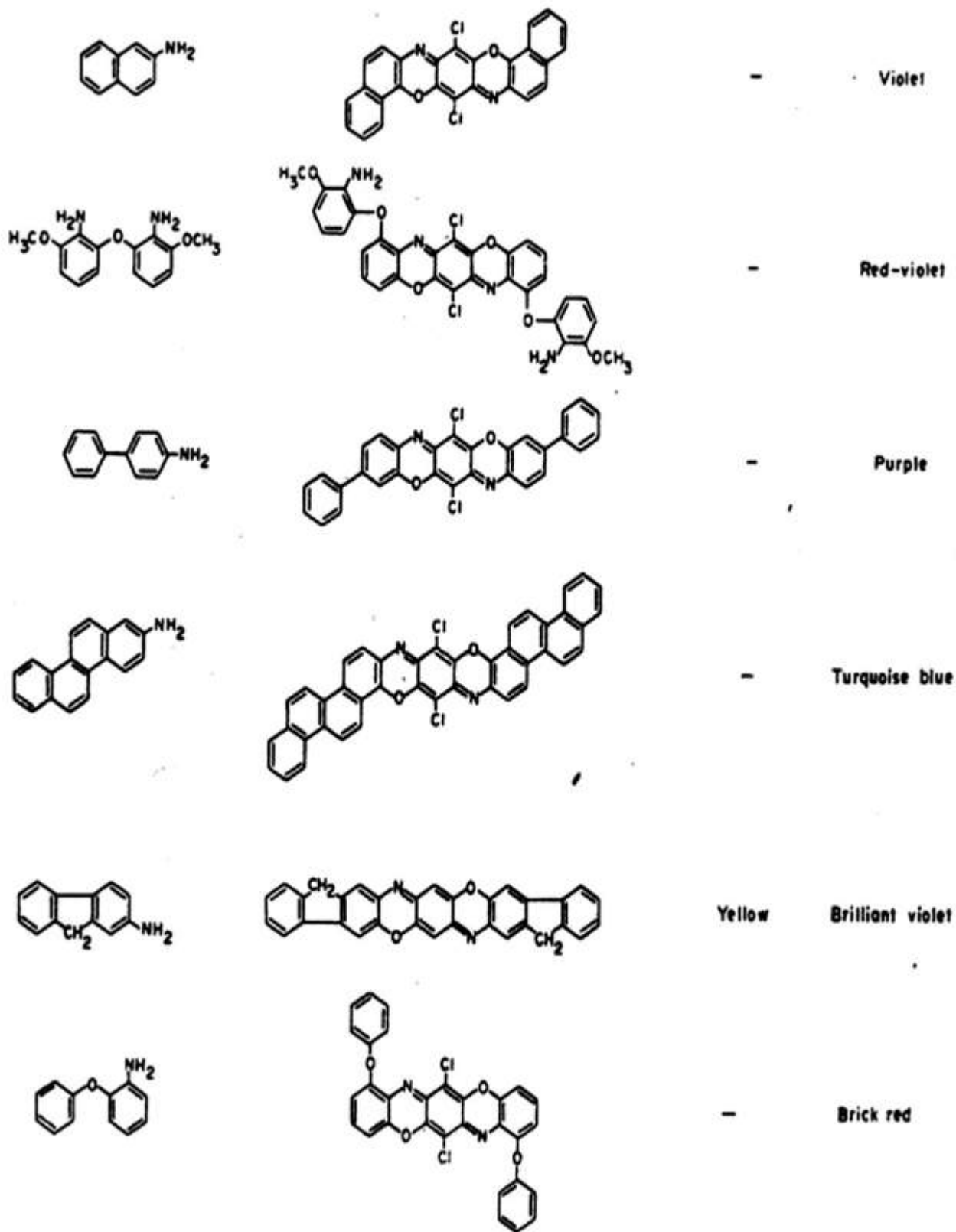
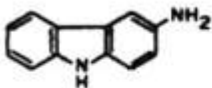
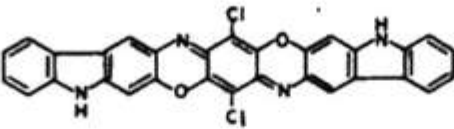
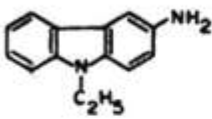
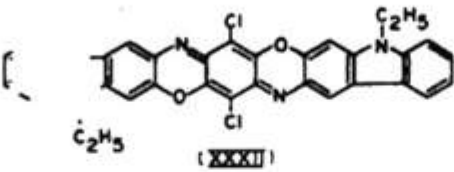
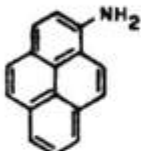
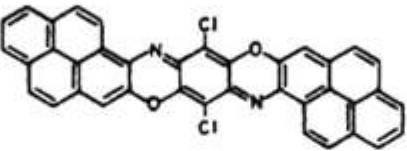
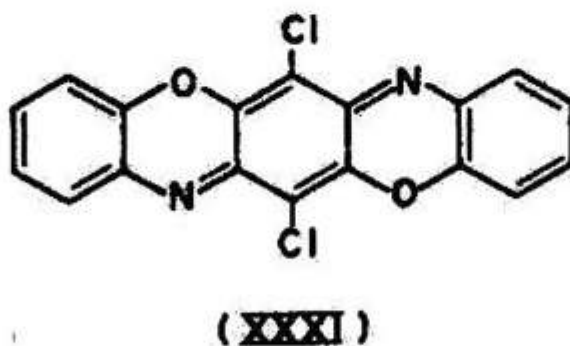


TABLE I—continued

<i>Amine</i>	<i>Dioxazine</i>	<i>Color of vat</i>	<i>Shade on cotton</i>
		—	Violet-blue
		—	Blue
		—	Dark blue

reduction of their sulfonyl chloride derivatives to the thiols. Condensation of chloranil with aniline gives 6,13-dichlorotriphenodioxazine (XXXI), which is chloro-sulfonated at 130°-135°. The chloro-sulfonation mass is drowned in ice and reduced with iron at 40°-55°



to give a bronzy powder which dyes cotton from a green-blue sulfide vat in red to claret shades of good chlorine fastness. Other dioxazines may be treated similarly to yield sulfur dyes of varying shades. The starting amine, dioxazine intermediate, and the shade of dye obtained are listed in Table I.

The dioxazine (XXXII, see Table I) (CSD II, 786) yields navy blue sulfur dyes by treatment of its sulfonyl chloride or sulfinate with thiourea.

G. OTHER INTERMEDIATES

Brown, khaki, and olive-green dyes are obtained by baking naphthols, nitroso-naphthols, and nitroso-phenols with a copper salt and poly-sulfide. The intermediates and shades of the resulting dyes are listed in Table II.

TABLE II
INTERMEDIATES FOR DYES MADE BY THIONATION

Intermediate	Thiolation temperature	Shade
2-Naphthol	255 ^o -260 ^o	Khaki
1-Nitroso-2-naphthol	260 ^o -265 ^o	Brownish khaki
p-Nitroso-phenol	260 ^o -265 ^o	Olive-green
1,5-Dihydroxynaphthalene	260 ^o -265 ^o	Brown

Under similar conditions phenolphthalein and fluorescein yield olive-green dyes.

p-Dichlorobenzene recovered from DDT waste is nitrated and then hydrolyzed to give 4-chloro-2-nitrophenol. Reflux thionation of this intermediate in aqueous polysulfide gives a gray sulfur dye.

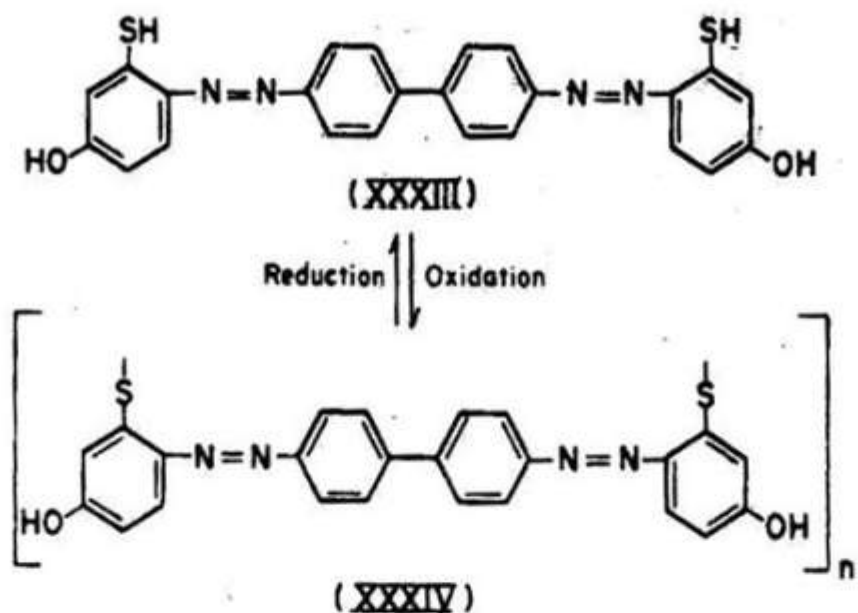
V. Sulfur Dyes of Known Constitution

As early as 1922 there were attempts to synthesize dyes of definite structure, which would be applicable to cotton-like sulfur dyes. Azo dyes were made having mercapto substituents, and these were soluble in sulfide and could be oxidized on the fiber to the disulfide. Benzidine was tetrazotized and coupled with 2 moles of m-mercaptophenol to give the dye (XXXIII). This dissolved in sodium sulfide to give a red solution which dyed cotton, and upon oxidation the polydisulfide form of the dye (XXXIV) was generated. This



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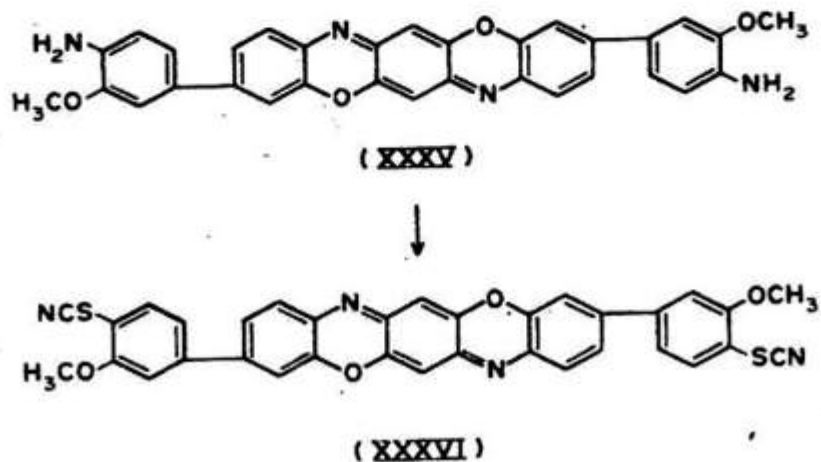
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and other azodisulfide dyes had certain technical defects, and no useful synthetic sulfur dye was developed until the 1940s when red and green sulfur dyes were invented, based on perylene-3,4,9,10-carboxylic diphenylimide and copper phthalocyanine, respectively. The commercial success of the green dye has stimulated research in other areas, including azo and dioxazine dyes.


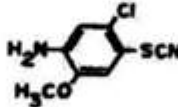
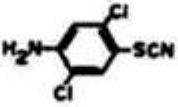
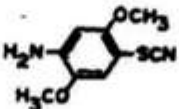
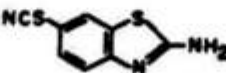
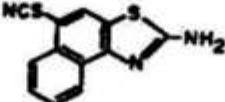
A. DIOXAZINES

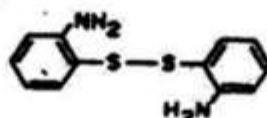
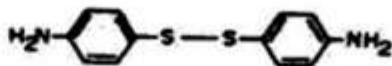

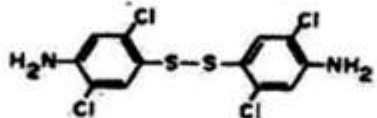
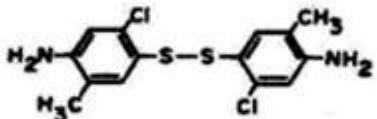
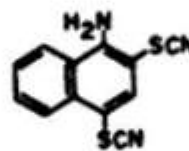
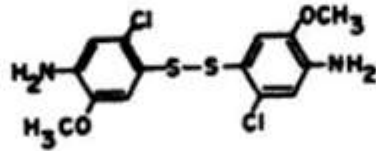
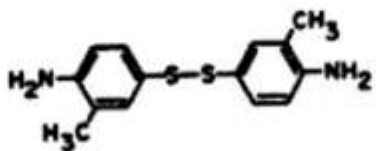
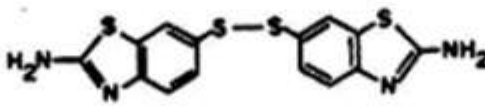
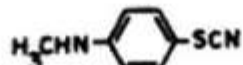
Chloranil is condensed with 2 moles of dianisidine to give the diamino-dioxazine (XXXV). This is tetrazotized and the tetrazonium salt is treated with sodium thiocyanate to yield the violet sulfur dye (XXXVI),


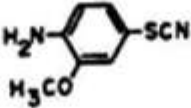
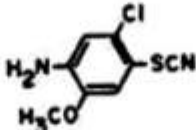
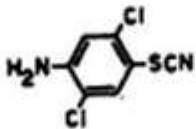
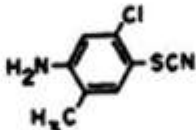
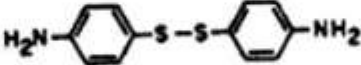

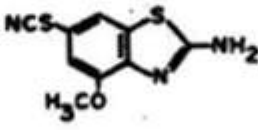
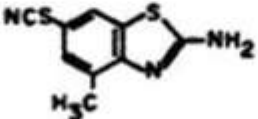


Condensation of dioxazine dicarboxylic acids with sulfur-substituted anilines yields orange to red sulfur dyes. For example, 6,13-dichloro-triphenodioxazine-2,9-dicarboxylic acid (XXXVII) is heated under reflux in o-dichlorobenzene with thionyl chloride and a catalytic quantity of pyridine to form the acid chloride (XXXVIII). The excess thionyl chloride is distilled off and 4-thiocyanatoaniline is added. Heating for 1 hour at 125° completes the reaction, and the dye (XXXIX) is isolated by diluting the reaction mixture with alcohol and filtering. Other dioxazines which may be used are 6,13-dichlorotripheno-dioxazine-3,10-dicarboxylic acid (XL), 3,6,10,13-tetrachlorotripheno-dioxazine-2,9-dicarboxylic acid (XLI) and 6,13-dibromotriphenodioxazine-3,10-dicarboxylic acid (XLII). Table III gives the shades obtained from these intermediates with various sulfur-containing amines.

TABLE III
DIOXAZINE DYES

<i>Dioxazine</i>	<i>Amine</i>	<i>Shade</i>
(XXXVII)		Red
(XXXVII)		Red
(XXXVII)		Yellowish red
(XXXVII)		Bluish red
(XXXVII)		Bluish red
(XXXVII)		Bluish red

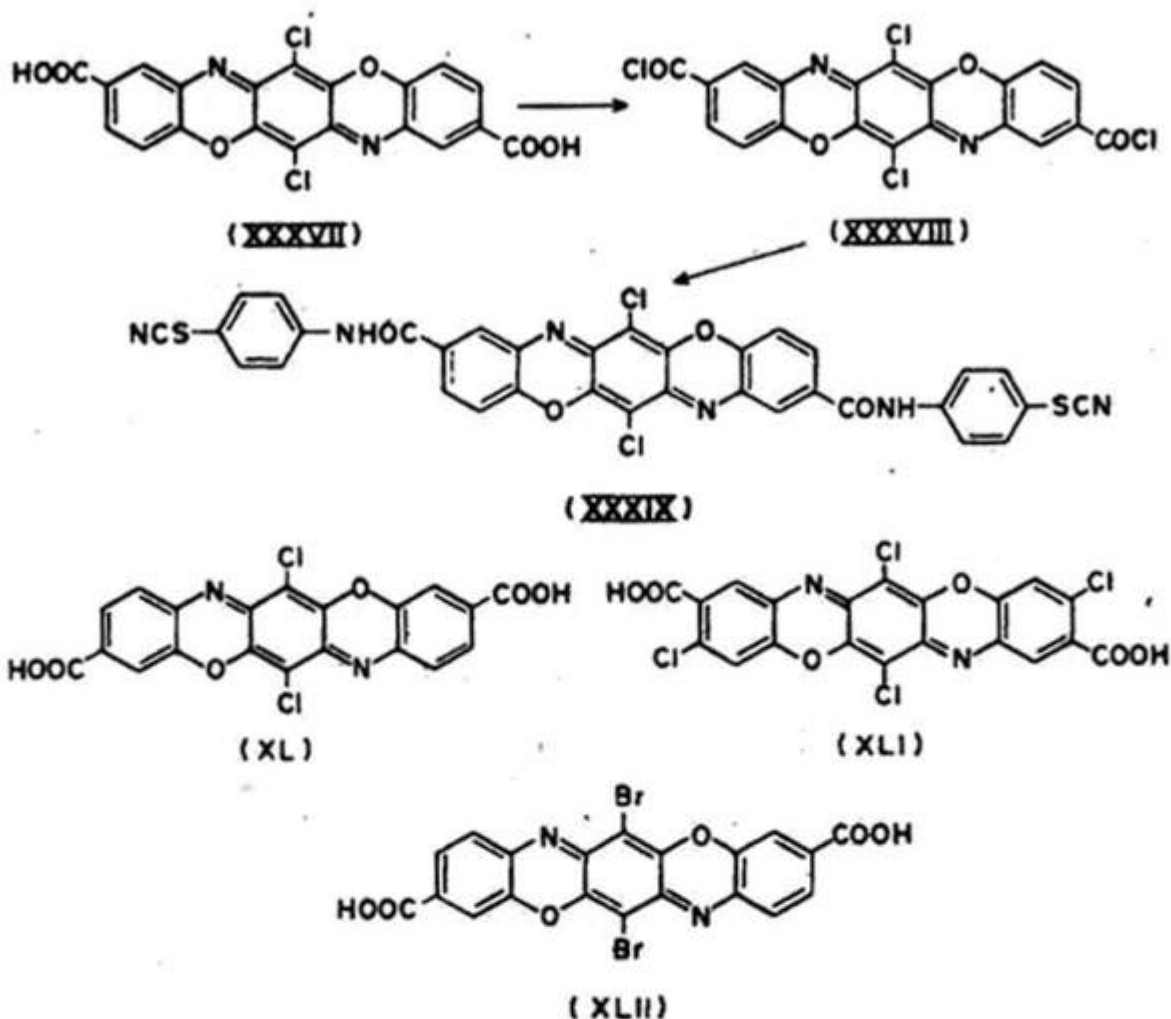
(XXXVII)		Red
(XXXVII)		Red
(XLII)		Orange
(XXXVII)		Yellowish red
(XXXVII)		Red
(XXXVII)		Bluish red
(XXXVII)		Red
(XXXVII)		Yellowish red
(XXXVII)		Bluish red
(XXXVII)		Light red

(XL)		Bright orange
(XL)		Bright orange
(XL)		Bright orange
(XL)		Yellowish orange
(XL)		Yellowish orange
(XL)		Bright orange
(XL1)		Red
(XXXVII)		Bluish red
(XXXVII)		Bluish red

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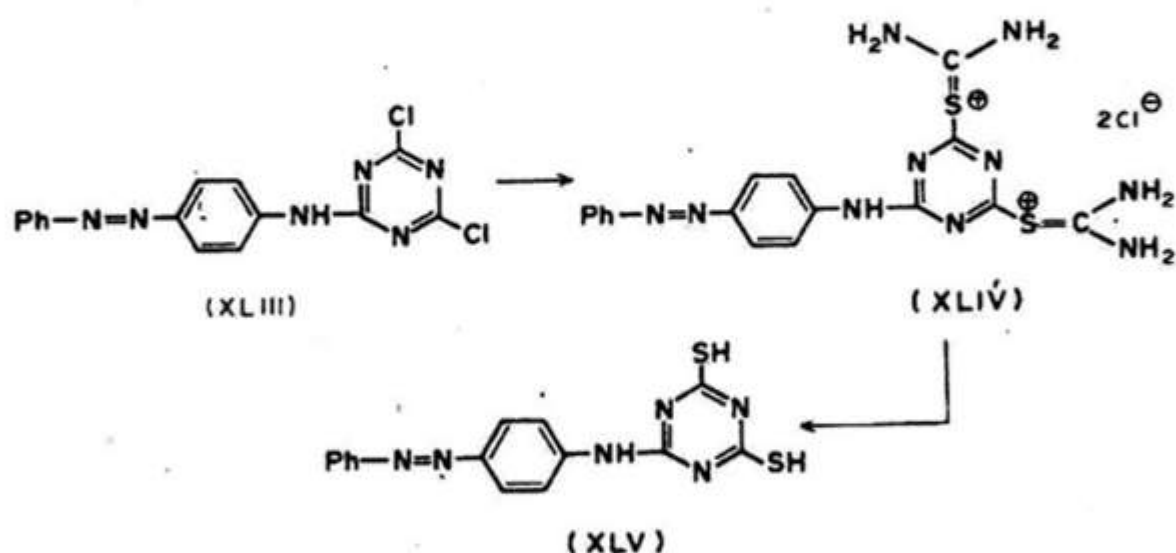


B. HIYAMAN DYES

In a 1959 monograph Hiyama reviewed the chemistry of synthetic sulfur dyes and proposed a theory to account for the dullness of sulfur dyes. It is held that sulfur dyes are dull in shade not merely because they are mixtures, but because mercaptan groups added to a chromophore widen the absorption band. Supporting evidence is found in the ultraviolet absorption spectra of certain colorless aromatic mercaptans, sulfides, and disulfides. The bright shades of certain sulfur yellows and blues are explained as being due to the dye's absorption band being only partially within the visible region. In order to synthesize bright sulfur dyes, it was considered necessary to separate the mercaptan groups from the chromophore by an insulating group which would prevent the electronic interaction that causes the absorption region to broaden out. Suitable insulating groups

include sym-triazine and sulfonanilide groups.

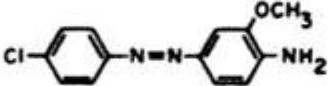
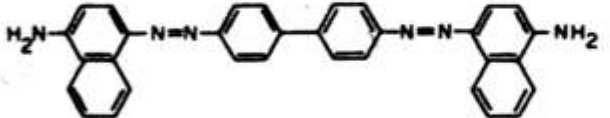
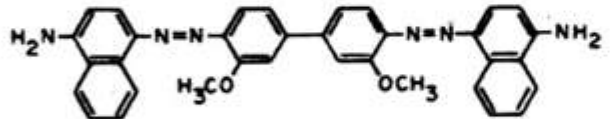
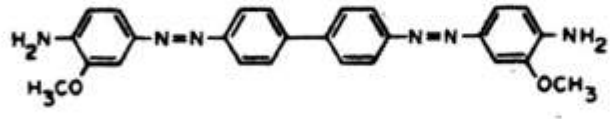
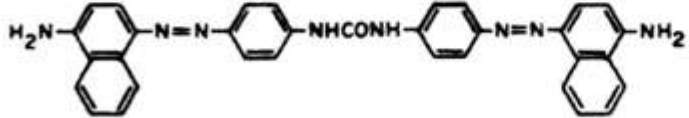
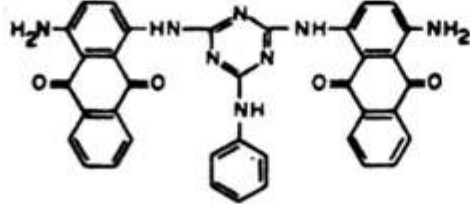
Cyanuric chloride is condensed with water-insoluble dyes having one or more primary amino groups. The remaining chlorine atoms on the triazine ring are then replaced by reaction with thiourea, and subsequent reduction of the isothiurea group gives the dimercaptotriazinyl-substituted dye. Thus aminoazobenzene is condensed with cyanuric chloride in acetone below 5°, using sodium hydroxide as an acid-binding agent. The dichlorotriazinyl intermediate (XLIII) reacts with 2 moles of thiourea at 10°-20° in the same solvent. Sodium sulfide is added and the thiourea derivative (XLIV) is reduced at 40°-50° to give the dye (XLV). The acetone is distilled off and the yellow dye is precipitated by acidifying with hydrochloric acid. The shades obtained from various other amino-substituted dyes are illustrated in Table IV.



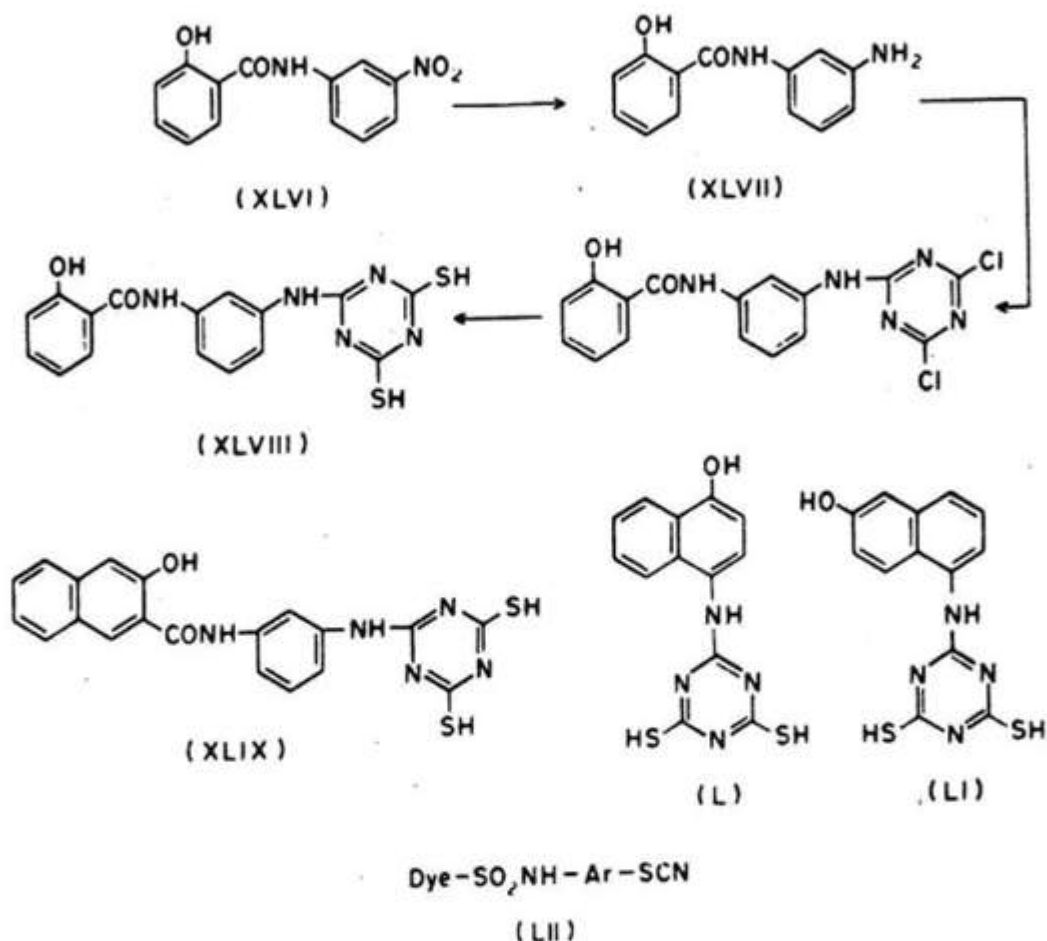
Other dimercaptotriazinyl azo dyes are made by coupling diazonium salts with coupling components carrying the dimercaptotriazinyl group. Salicylic acid and m-nitroaniline are heated under reflux with phosphorus trichloride in toluene to form the m-nitroanilide (XLVI) in 85% crude yield. Iron reduction in dilute acetic acid gives (XLVII) in 80% yield. This is condensed with cyanuric chloride in acetone and subsequently treated with thiourea to give (XLVIII). The coupler (XLIX) is made from Naphthol AS-BS by sulfide reduction followed by condensation with cyanuric chloride, then thiourea, while (L) and (LI) are made analogously from the corresponding aminonaphthols. A wide variety of shades (from

yellow to blue) has been produced by coupling these four intermediates with various diazonium salts. One of the technical disadvantages of these dyes is that mercaptan groups on the triazinyl ring are rather more difficult to oxidize than are aromatic thiols. Consequently the oxidation of the dyeings is incomplete, and the residual thiol groups give rise to inferior wash-fastness. This can be improved by treating the dyeing with a salt of a heavy metal, such as cadmium, or with an alkylating agent such as benzyl-di-methyl-phenyl-ammonium chloride.

TABLE IV
DYES CONTAINING THE DIMERCAPTOTRIAZINYL GROUP

<i>Amine</i>	<i>Shade of derived dye</i>
	Yellow
	Brown
	Brown
	Orange yellow
	Yellow orange
	Violet

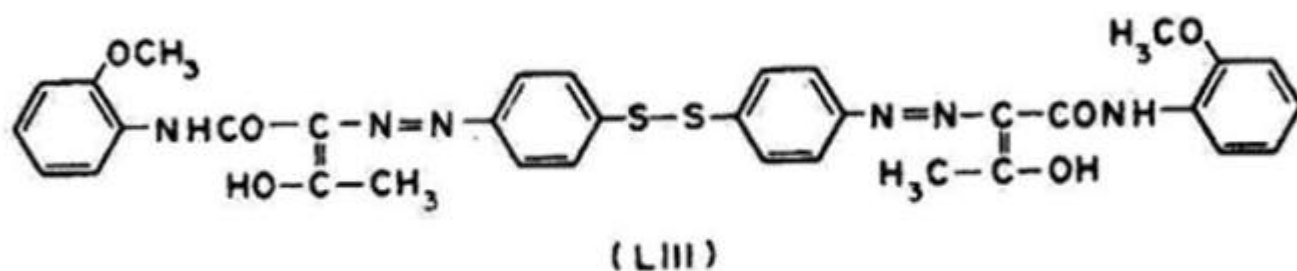
A number of dyes are described in which an azo or phthalocyanine pigment is chloro-sulfonated and the sulfonyl chloride condensed with a diamine. The free amine group is diazotized and converted to thio cyanate. The products are dyes of general formula (LII). Their wash fastness is not good, partially because of the acidic nature of the sulfonamide group, which makes the dyes somewhat soluble in alkali.



C. AZO DISULFIDE DYES

More recently azo disulfide dyes have been made in which the mercapto groups are not separated by insulating groups from the chromophore. Many of these dyes give bright shades and have technically satisfactory wash-fastness. Bright greenish yellow shades of good wash-fastness are obtained by coupling various diaminodiaryl disulfides with acetoacetanilides. For example, 4,4'-diaminodiphenyl disulfide is tetrazotized in dilute hydrochloric acid at 0°, and coupled with a solution of acetoacet-o-anisidine in dilute

caustic soda. The product (LIII) is filtered off, washed, and dried. Improved lightfastness is obtained by using 4,4'-diamino-2,2'-dichlorodiphenyl disulfide as the diazo component. Coupling with pyrazolones gives yellow dyes, while derivatives of Naphtol AS give reds. These azo disulfide dyes all have rather low affinity for cellulose and a special technique has been developed for their application. The dyes are prepared in dispersion form and are padded on to the cloth, which is then dried. Reducing agent is then padded on and the cloth is steamed to fix the dye. After washing, the cloth passes through a sodium dichromate/acetic acid bath to oxidize the dye. The dyeing is then washed in water and soaped. Azo disulfide dyes may be applied by this method along with sulfur dyes, either as dispersed pastes, reduced liquids, or their thiosulfate derivatives. Instead of the above two-pad procedure the dye may be padded along with a compound such as thiourea or di(sodiocarboxymethyl)trithio-carbonate which has no effect on the dye in the pad box, but reduces it in the steamer.



To be Continue in next Issue.....



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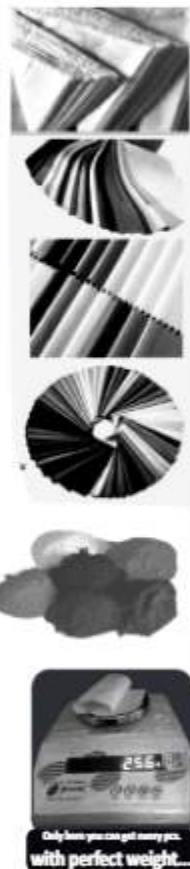
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APPLICATION

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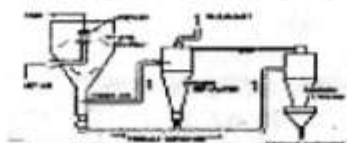
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- Carmosine
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- **SOLVENT RED 23**
- **SOLVENT RED 24**
- **SOLVENT YELLOW 14**
- **SOLVENT YELLOW 2**
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Email : emco@emcochemicals.com | Website : www.emcochemicals.com

CIN No. : U24110 Mh1990 PTC056053 | GST IN : 27AAACE1167D122

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INDUSTRIAL CHEMICALS

<ul style="list-style-type: none"> • 1-Bromo-3-Chloropropane • 1,4 Dihydroxy Anthraquinone • 2,2 Dimethoxy Propane • 2,3,2,4 Xylidine • 2,5 Dichloro Aniline • 2-Ethyl Hexanoic Acid • 2-Ethyl Hexanol • 2-Ethyl Hexyl Acrylate • 2-Ethyl Hexyl Methacrylate • 2-Methyl THF • 2-Methyl-5-Nitro Aniline • Acetonitrile • Acrylamide • Acrylic Acid • Adipic Acid • Alpha Naphthol • Ammonium Thiocyanate • Anhydrous Aluminium Chloride • Aniline • Anthraquinone • Beta Naphthol • Benzoic Acid • Benzophenone • Biophenol A • Butyl Acetate • Butyl Cellosolve • Caustic Potash 	<ul style="list-style-type: none"> • Caustic Soda Flakes • Cyanoacetamide • Cyanuric Chloride • Cyclohexylamine • Di-isopropyl Ether • Dicyandiamide • Diethanolamine • Diethyl Malonate • Diethylene Triamine • Dimethyl Acetamide • Dimethyl Aino Propyl Amine • Dimethyl Carbonate • Dimethyl Formamide • Dimethyl Malonate • Di-N-Butyl Amine • Diphenyl Amine • EDTA 2Na / 4 Na / Free Acid • Epichlorohydrin • Ethyl Chloro Formate • Ethyl Cyano Acetate • Ethylene Diamine • Formic Acid 85% / 99% • Furfuryl Alcohol • Glyoxal 40% • HEDP 60% • Hexaethylene Pentamine (HEPA) 	<ul style="list-style-type: none"> • Hydrobromic Acid 48% • Hydroquinone • Hydroxylamine Hydrochloride • Isobutanol • Isophthalic Acid • Isopropyl Alcohol • Isovaleraldehyde • Isophorone • Isophorone Diamine • L(+)/ D / Di Tartaric Acid • Lithium Hydroxide • Maleic Anhydride • Melamine • Meta Amino Phenol • Meta Chloro Aniline • Methane Sulphonyl Chloride • Methyl Aceto Acetate • Methyl Chloro Formate • Methyl Cyano Acetate • Methylene Dichloride • Methacrylic Acid • Mono Chloro Benzene • Mono-N-Butyl Amine • Monoethanolamine • Morpholine • N-N-Dimethyl Aniline • Nitrobenzene 	<ul style="list-style-type: none"> • N-Butanol • Neo Pentyl Glycol • N-Hexane • N-Methyl Piperazine • N-Propanol • Ortho Amino Phenol • Ortho Chloro Benzoic Acid • Ortho Chloro Benzaldehyde • Ortho Chloro Benzene • Para Anisidine • Para Chloro Benzaldehyde • Para Hydroxy Benzaldehyde • Para Hydroxy Benzoic Acid • Para Nitro Benzoic Acid • Para Toluic Acid • Paraformaldehyde (96%, 91%) • Parachloroethylene • Phosphoric Acid Food Grade 85% • Phosphorous Pentoxide • Piperazine Anhydrous • Potassium Carbonate Grannules • Potassium Carbonate Powder • Propylene Glycol • Propionaldehyde • Phthalic Anhydride 	<ul style="list-style-type: none"> • Resorcinol • Refined Glycerine • Salicylic Acid (Tech. Grade) • Sodium Borohydride • Sodium Hexa Meta Phosphate • Sodium Meta Bisulphite • Sodium Nitrite • Sodium Per Carbonate • Sodium Sulphide 50% Yel Flake • Sodium Gluconate • Sodium Methoxide • Stearoc Acid • Succinic Acid • Sulphanilic Acid • Tertiary Butanol • tetrahydrofuran • Thionyl Chloride • Thiophenol • Thiourea • Tri Ethyl Amine • Tri Ethyl Ortho Formate • Tetraethylene Pentamine (IEPA) • Tri N Butyl Amine • Triethanolamine (85%) (99%) • Triphosgene
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DYES INTERMEDIATES

> Gamma Acid	> Aniline 2-4 Di Sulphonic Acid	> K. Acid
> Acetyl Bronners Acid	> Aniline 2,5 Di Sulphonic Acid	> R. Salt
> Bronners Acid	> G. Salt	> Benzaldehyde-2-Sulfonic Acid (OBSA)
> Schaeffers Acid	> Sulpho Tobias Acid (STA)	> Benzaldehyde 2,4 Disulphonic Acid

PIGMENT INTERMEDIATES

> MAA / AAMX / AAOA / AAOCA
> Bon Acid
> Naphthol AS / ASE / ASIRG / ASLC / Gum Rosin (WW)
> Tobias Acid

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Email : emco@emcochemicals.com

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| ● ACETANILIDE | ● O.D C. B (INDIAN IMPORTED) |
| ● ACRYLIC ACID | ● PARA TOLUIDINE |
| ● ALPHA NAPHTHOL | ● SALYCILIC ACID |
| ● ACRYLAMIDE | ● THIOUREA |
| ● ANHRAQUINONE | ● UREA TECH. (UAE) |
| ● BETA NAPHTHOL | ● NAPHTHALENE CRUDE/ REFINED |
| ● BENZALDEHYDE | ● SODIUM NAPHTHIONATE |
| ● BENZOYL CHLORIDE 99.5% | ● SODA ASH (IMPORTED) |
| ● BENZOIC ACID (CHIINA) | ● CYANURIC CHLORIDE |
| ● EDTA | ● BENZOYL PEROXIDE |
| ● GUM ROSIN (INDONESIA) | ● CYNURIC CHLORIDE |
| ● MONO/DI/TRI/ETHANOLAMINE | ● O P D A |
| ● MONO ETHYL/METHYL ANILINE | ● RESOURCINOL |
| ● DIETHYL/METHYL ANILINE (DEA*DMA) | ● GLYCERINE REFINED |
| ● 2NADAPSA | ● DCDA(DYNDIAMIDE) |
| ● GARNET G, B 99% (O.A.A.T) | ● SODIUM MOLYBDATE |
| ● MELAMINE (CHINA) | ● POTASSIUM PERMAGANATE |
| ● ORTHO/PARA ANISIDINE | ● PAAB |

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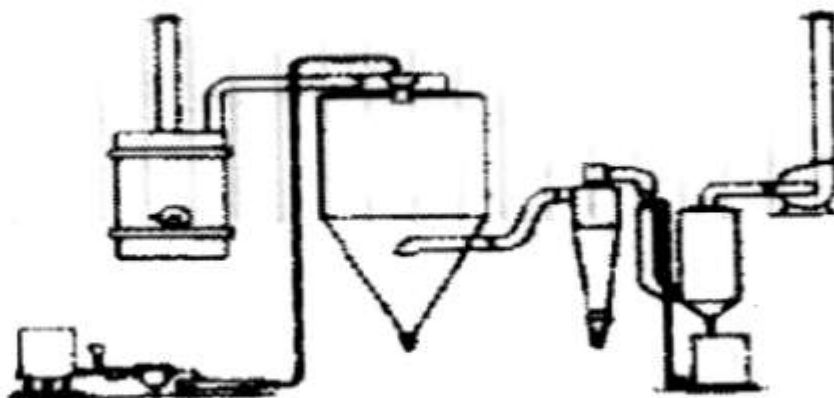
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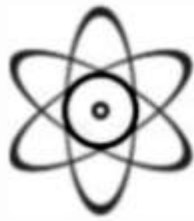
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146, 254
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PIGMENT PASTE

- ◆ FIG. YELLOW GG
- ◆ FIG. GOL. YELLOW RM
- ◆ FIG. ORANGE G
- ◆ FIG. GREEN G
- ◆ FIG. RED GR
- ◆ FIG. VIOLET B
- ◆ FIG. BLUE BR
- ◆ FIG. N. BLUE B
- ◆ FIG. T. BLUE BN
- ◆ FIG. BROWN GR
- ◆ FIG. BLACK NG
- ◆ FIG. RED VIOLET B
- ◆ FIG. BORDEAUX R
- ◆ FIG. PINK BBJ

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- ◆ SYNTHETIC
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● Gamma Acid	● BNSA (2 Naphthalene Sulphonic Acid)
● STA (Sulpho Tobias Acid)	● Benzaldehyde 2,4 Disulphonic Acid
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- > SACTEX CORROSAFE-TQ (Corrosion inhibitor for water circulatory System)
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- > SACTEX-SCOURALL (Neutral cleaner concentrate)

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E-mail : info@marksundyed.com

Web : www.marksundyed.com

Contact Person : MR. VIKAS KEDIA (M) +91 - 93755 22206

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Contact Address:

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(M) 098251 75508, 098254 69245 E-mail ID : pbtradingco17@yahoo.co.in, tanaydyechem@yahoo.com

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Manufacturer & Exporter of :

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- Reactive Orange ME2RL
- Reactive Red ME4BL
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- Reactive Blue RGB
- Reactive Blue ME2GL
- Reactive Blue ME2RL

V.S. Dyes

- Reactive Black B
- Reactive Black WNN
- Reactive Black GDN
- Reactive Black NCL

Intermediates

- BHK Acid

Solvent Dyes

- Solvent Orange 3

Printing Dyes

- Reactive Yellow PR
- Reactive Yellow FG
- Reactive Yellow P6G
- Reactive Orange P2R
- Reactive Red P8B
- Reactive Red P2B
- Reactive Red PB
- Reactive Magenta PB
- Reactive Navy Blue PX
- Reactive Blue P3R

Acid Dyes

- Acid Black 210
- Acid Black 1
- Acid Black 194
- Acid Yellow 36
- Acid Red 73
- Acid Blue 9

New Product

- Reactive Yellow CD
- Reactive Orange CD
- Reactive Red CD
- Reactive Yellow JJ
- Reactive Red JJ
- Reactive Blue BF
- Reactive Blue BB
- Reactive Red BB
- Reactive Red MEGF

Direct Dyes (N.B.)

- Direct Black EX / BH
- Direct Green B
- Direct Bordeaux BX
- Direct Blue FF
- Direct Brown MR
- Direct Red 12B / 4B5
- Direct Orange S | Yellow 44
- Direct Helio B
- Direct Congo Red
- Direct Violet MB | Rose FR

Plot No. : 5634/I, Phase-II, G.I.D.C. Estate, Opp. Green Environment, Vatva, Ahmedabad - 382 445. (Gujarat - INDIA).

Admn. Office : 501, Abhishree Adroid, Nr. Swaminarayan Temple, Mansi Circle, Vastrapur, Ahmedabad - 380009 (Gujarat - INDIA). Tel. : +91-79-48002201 / 02

Contact Person - Mr. Darshan Shah (M) +91-9825041016 / 90999 48640

E-mail : nemichem@hotmail.com | nemi2323@gmail.com

Website : www.nemichem.com

ANKUR DESAI

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DIRECT DYES

ACID DYES

DYES INTERMEDIATES

E-mail : ankurdesai42@yahoo.co.in

Mobile : 098254 46476

OFFERS : EX- WAREHOUSE AHMEDABAD DUTY PAID & EX-BOND

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- BROMAMINE ACID
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Please Contact :

DOUBLE 
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Prahlanagar,
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Tel. :- +91- 79 - 40321887,

Fax : +91 - 79 - 40321881

M : +91- 99250 10682

E-mail : double@doublechem.com,
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Dist. Rajkot, (Gujarat). India.

✉ avsarchloratech@gmail.com
🌐 www.avsarchloratech.com

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DIRECT DYES

- GREEN B (NB)
- GREEN PLS (NB)
- BLACK E (NB)



Bhavin ENTERPRISE

Plot No. 1032/18, Phase II, G.I.D.C. Panoli - 394 116. Ankleshwar,
Dist. Bharuch (Gujarat) Email : bhavinenter@gmail.com

Contact Person : SADUL BHAI (M) 098258 35457 / BHAVIN BHAI (M) 07228990060



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- ▶ VIOLET 2R
- ▶ BLUE 4B
- ▶ BLUE 4G
- ▶ GREEN FFB
- ▶ BROWN RRD
- ▶ GREY 3B
- ▶ BLACK BG

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- ▶ YELLOW 2G
- ▶ GOL. YELLOW RM
- ▶ ORANGE 5G
- ▶ RED GR
- ▶ PINK B
- ▶ ROSE PINK
- ▶ RED VIOLET 6B
- ▶ RUBINE B
- ▶ BORDEAUX
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E-mail : kemcolproduct@yahoo.com, kemcol053@gmail.com

Contact Person : Mr. Arun Jindal (M) +91 - 93270 24805 , 76980 01984



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Manufacturer & Exporter of Dyestuffs

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• BLACK 71	GREY BLACK W
• BLACK 84	GREY GC
• BLACK 194	BLACK MSRL
• BLACK 52	BLACK WA
• BLACK 210	BLACK NT
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• BLUE 113	N. BLUE 5R
• VIOLET 90	BORDEAUX MB
• GREEN 73	OLIVE GREEN S-G
• BROWN 14	BROWN RD
• BROWN 58	BROWN DS
• BROWN 75	BROWN NR
• BROWN 83	BROWN CGG
• BROWN 98	OLIVE BROWN G

• BROWN 106	BROWN RL
• BROWN 121	BROWN DH
• BROWN 122	DARK BROWN R
• BROWN 161	BROWN NGB
• BROWN 188	BROWN F3G
• BROWN 214	BROWN RT
• BROWN 282	BROWN S-GR
• BROWN 348	BROWN ESR
• BROWN 349	BROWN ESG
• BROWN 354	BROWN HBR
• BROWN 355	BROWN M-BL
• BROWN 365	BROWN M-2RL
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HOT SERIES

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• Blue 201	SUPRA BLUE BRL
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• Red 239	RED 2G
• Black 80	BLACK OB

C-1, 257/2, Phase II ,G.I.D.C., Vatva, Ahmedabad - 382445 (Gujarat-INDIA)

• Email : hinadye_chem@yahoo.in • Website : www.hinadyechem.com

Contact Person :-

MR. SIDDHARTH PATEL (M) +91-9879103924

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Mobile : +91 - 98251 60650, 98256 35952
E-mail : info@etaorganics.com , dassudhamalia@gmail.com
Website : www.etaorganics.com

Manufacturers & Exporters of Dyestuffs

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- | | |
|-----------------------------|--------------------------|
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| > Red P2B (Red-45) | > Purple (Mix) |
| > Red P2BP (Red 3:1) | > Magenta MEBL (---) |
| > Red P4BN (Red-245) | > Yellow W3R (---) |
| > Red P6BN (Red-218) | > Orange W3R (---) |
| > Red P8B (Red-31) | > Red W3R (---) |
| > Red ME3BL (Red-194) | > Blue ME2GL (Blue 194) |
| > Red ME6BL (Red-196) | > Blue BF (Blue 222) |
| > G. Yellow HR (Orange.-12) | > Ultra Navy Blue RGB |
| > G. Yellow RR (---) | |
| > Yellow MERL (Yellow 145) | |

Reactive R.O. Dyes

- 5 Red HE3B (Red-120)
- 5 Red ME4BL (Red-195)
- 5 Red BS (Red-111)
- 5 Red CD (---)
- 5 Orange CD (---)
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- 5 Blue H5R (Blue 13)
- 5 Brown P4R (Brown-11)



(An ISO 9001 : 2008 CERTIFIED COMPANY)

C-1 / 423, Phase-II, G.I.D.C., Vatva,
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Tel. : +91 - 079 - 4008 3369
Mobile : 098252 34676 / 09638865713
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- ★ ACID GREEN V (Green 16)
- ★ ACID VIOLET 5BS (Violet 17)
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**Also Available
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E-mail : info@nirvanaenterprise.in

Contact : MR. MANAN PATEL (M) +91 - 90999 74010

Manufacturer of :

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- Re. Orange 2R (Re. Orange 7)

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(Orange BRO)
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Email : setracolorchem@gmail.com

Contact : Mr. Avam Jagnani (M) +91 - 97122 11444 / Mr. Nawal Jagnani (M) +91 - 97122 00077

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| • ORANGE TGLL (Or. 39) | • RED 12B (R.31) | |



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Plot No. 316, Opp. New Maruti Way Bridge, Phase-II, G.ID.C., Vatva, Ahmedabad-382445

E-mail : patelshail_1978@yahoo.com

Contact : Mr. Ketan Patel (M) 088664 88472 / Mr. Shailesh Patel (M) 095583 51028

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Shed No. A-2/2214, B/h, E.T.L., G.I.D.C.Estate,
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E-mail : advancedyechemindustries@yahoo.com
info.adci@yahoo.com

Contact : MR. RAVI MAKANI (M) +91 94096 14706
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- RE. RED HE8B (RED - 152)
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- RE. ORANGE HE2R (ORANGE - 94)
- RE. G.Y. MERL (YELLOW - 145)
- RE. RED ME4BL (RED - 195)
- RE. RED ME6BL (RED - 250)
- RE. ORANGE ME2RL (ORANGE - 122)
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• E-mail : umangvyas8@gmail.com • Web : www.bptraders.in

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• Contact : Mr. Mohsin Richniwala (M.) +91 99784 41718

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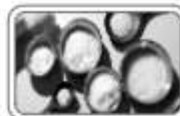
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Contact Person : MR. GIRISH PATEL (M) +91 - 93270 94416

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L-50, G.I.D.C., B/h. Old Water Tank, Opp. Fire Station, Odhav, Ahmedabad - 382 415. (Gujarat) INDIA.
E-mail :- siddhisurichem@rediffmail.com

Contact : Mr. Hetal Shah (M) 098795 45966 / Mr. Parag Shah (M) 094263 25551

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- MJ-P10 : Effluent treatment
- MJ-DS20 : Antiscalant for MEE

TEXTILE

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- MJ-Dpaste : Scouring Agent for Wool
Washing Agent for dyed & Printing textile
- MJ-Dyfix : Dye fixing Agent for Reactive & Direct Dyes
- MJ-66 : Levelling Agent and Brightening Agent for Pigment Printing
- MJ FC : Stain removing cum Scouring Agent
- MJ F10 : Effluent Treatment
- MJ 300 : Mercerising Agent

FERTILIZER : Defoamer for fertilizer / Wetting Agent / Penetrating / Sticker & Spreading Agent

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PAINTS : Wetting Agent : Wall Putty

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- MJ Plast : Plasticizer for R.C.C.



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Email : hl.enter@yahoo.in , info@shreeuma.com

Website : www.hlenter.in, www.shreeuma.com

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S.S. 316 GMP Reactor



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30KL S.S. 316 with MS Limped Coil Vessel



10KL M.S. Sulphonator



S.S. 316 Reactor Vessel



S.S. 316 Pilot Plant Vessel



M.S. Ball Mill



MS Velkenizer Bhattho



Heat Exchanger



S.S. 316 Storage Tank



Chemical Grinder Machine



Ice-Crusher



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- Pig. Red-2103 (Red-3)
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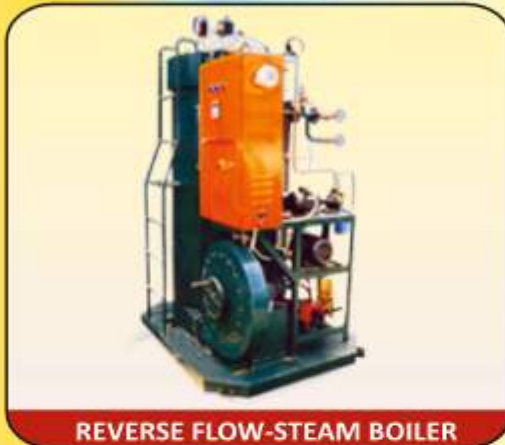
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- Acid Brown 83
- Acid Brown 97
- Acid Brown 98
- Acid Brown 106
- Acid Brown 147
- Acid Brown 161
- Acid Brown 165
- Acid Brown 188
- Acid Brown 348
- Acid Brown 354
- Acid Brown 355
- Acid Brown 365
- Acid Brown 425
- Acid Brown 432
- Acid Brown 434
- Acid Brown 452
- Acid Green 68:1
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- Disperse Tomato Red LT
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- Dis. Acetate Yellow 3 H/c
- Dis. Acetate Orange 1 H/c
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- SFD Brown GTLC
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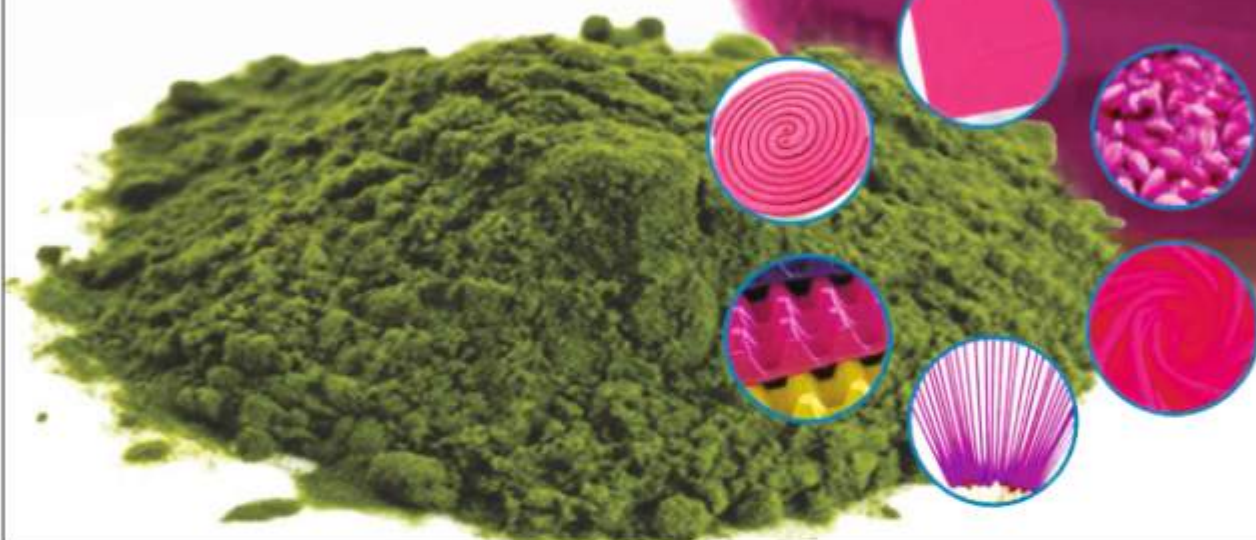
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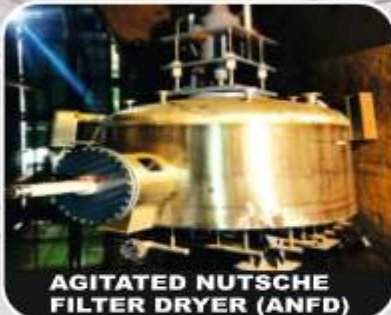
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N,N-Diethyl Meta Amino Acetanilide (S-12) 6375-46-8 N,N Diethyl Meta Amino Propionilide (M-9) 22185-75-7

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Leuco-1,4-Diaminoanthraquinone 476-60-8 Meta Uredo Aniline (MUA) 0103-03-07

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- Yellow M4G (Y.22)
- Yellow P6G (Y.95)
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- G. Yellow M4R (O.14)
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- Red P2B (R.45)
- Red HE3B (R.120)
- Red H4BN (R.3:1)
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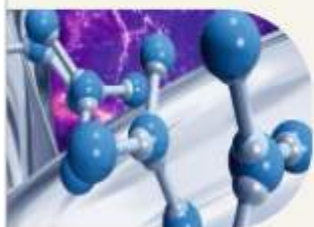
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- * Dt. Pink 3B Red 254
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- * A. Scarlet G (Red 97)
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- * A. Navy Blue 5R (Blue 113)

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Re. Red H4BN	(Red 3.1)
Re. Red PB	(Red 24)
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Re. Red XL7B	(Unknown)
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Re. Purple P3R	(Violet 1)

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Re. Red ME4BL	(Red 195)
Re. Red ME6BL	(Red 250)
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Re. G. Yellow MERL	(Yellow 145)
Re. Magenta MEB	(Unknown)

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| ✓ DT. YELLOW G (Y. 107) | ✓ DT. BLUE G (BL. 8) |
| ✓ DT. YELLOW 2G (Y. 132) | ✓ DT. SKY BLUE FF (BL. 15) |
| ✓ DT. YELLOW PG (Y. 142) | ✓ DT. BLUE NB (BL. 54) |
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| ✓ A. YELLOW 5GN (Y. 110) | ✓ A. NAVY BLUE SRL (BL. 335) |
| ✓ A. YELLOW RL (Y. 194) | ✓ A. GREEN 3B (GR. 25) |
| ✓ A. YELLOW 55RL (Y. 204) | ✓ A. OLIVE GREEN BGL (GR. 104) |
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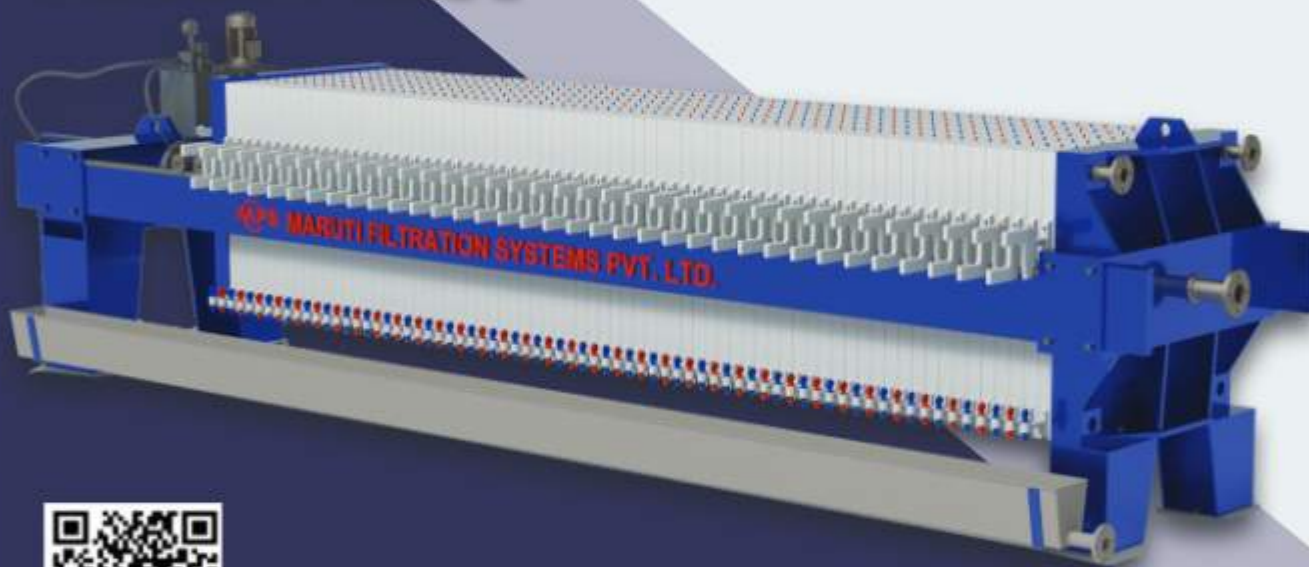
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